Rhizosphere to the atmosphere: contrasting methane pathways, fluxes and geochemical drivers across the terrestrial-aquatic wetland boundary

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Abstract

Although wetlands represent the largest natural source of atmospheric \( \text{CH}_4 \), large uncertainties remain regarding the global \( \text{CH}_4 \) flux. Wetland hydrological oscillations contribute to this uncertainty, dramatically altering wetland area, water table height, soil redox potentials and \( \text{CH}_4 \) emissions. This study compares both terrestrial and aquatic \( \text{CH}_4 \) fluxes over two distinct seasons in both permanent and seasonal remediated freshwater wetlands in subtropical Australia. We account for aquatic \( \text{CH}_4 \) diffusion and ebullition rates, and plant-mediated \( \text{CH}_4 \) fluxes from three distinct vegetation communities, thereby examining seasonal, diurnal and intra-habitat variability. \( \text{CH}_4 \) emission rates were related to underlying sediment geochemistry. For example, distinct negative relationships between Fe(III) and \( \text{SO}_4^{2-} \) and \( \text{CH}_4 \) fluxes were observed, whereas distinct positive trends occurred between \( \text{CH}_4 \) emissions and Fe(II)/AVS, where sediment Fe(III) and \( \text{SO}_4^{2-} \) were depleted. The highest \( \text{CH}_4 \) emissions of the seasonal wetland were measured during flooded conditions and always during daylight hours, which is consistent with soil redox potential and temperature being important co-drivers of \( \text{CH}_4 \) flux. The highest \( \text{CH}_4 \) fluxes were consistently emitted from the permanent wetland (1.5 to 10.5 mmol m\(^{-2}\) d\(^{-1}\)), followed by the Phragmites australis community within the seasonal wetland (0.8 to 2.3 mmol m\(^{-2}\) d\(^{-1}\)), whilst the lowest \( \text{CH}_4 \) fluxes came from a region of forested Juncus sp. (-0.01 to 0.1 mmol m\(^{-2}\) d\(^{-1}\)) which also corresponded with the highest sedimentary Fe(III) and \( \text{SO}_4^{2-} \). We suggest that wetland remediation strategies should consider geochemical profiles to help to mitigate excessive and unwanted methane emissions, especially during early system recovery periods.
1.0 Introduction

Wetlands are considered one of the most valuable ecosystems on Earth (Costanza et al., 2014). They are biodiversity hotspots that provide ecosystem services such as water filtration, sediment trapping, floodwater retention and carbon (C) storage (Bianchi, 2007). Wetlands account for ~5.5% of terrestrial surfaces (Melton et al., 2013) and have been estimated to store from ~4% (Bridgham et al., 2014) to ~30% (Mitsch et al., 2013) of Earth’s estimated 2500 Pg soil C pool (Lal, 2008). Pristine wetlands have long been considered net C sinks due to their high rates of productivity and low rates of decomposition (Mitsch et al., 2013); however due to their waterlogged nature and anaerobic soils, wetlands are ideal environments for the production of methane (CH$_4$), a potent greenhouse gas. As such, wetlands are recognised as Earth’s largest natural source of CH$_4$ to the atmosphere (185 ± 21 Tg C yr$^{-1}$) (Saunois et al., 2016).

Resolving the drivers, pathways and effects of seasonal weather oscillations on wetland CH$_4$ sink or source behaviours is important to enable more accurate climate model projections and to reduce uncertainties in the global wetland CH$_4$ budget (Kirschke et al., 2013; Saunois et al., 2016). Mitsch et al. (2013) estimated that the average ratio of freshwater wetland CO$_2$ sequestration to CH$_4$ emissions was 19.5:1. As CH$_4$ is 34 times more potent than carbon dioxide (CO$_2$) over a 100 year time scale (Stocker et al., 2013), this suggests that many freshwater wetlands may have a net positive radiative forcing effect on climate (Hernes et al., 2018). However, variability in geomorphology, wetland maturity, salinity and underlying geochemical composition all contribute to variable CH$_4$ dynamics (Bastviken et al., 2011; Poffenbarger et al., 2011; Whiting & Chanton, 2001). The lack of spatially-resolved wetland CH$_4$ emission data, as well as the limited number of studies constraining the multiple wetland CH$_4$ flux pathways (i.e. ebullition, diffusion and plant-mediated) coupled with ongoing anthropogenic conversion of wetland systems (Bartlett & Harriss, 1993; Neubauer & Megenigal, 2015; Saunois et al., 2016) further contribute to the uncertainties around CH$_4$ regional to global scale budgets.

Extensive clearing and drainage of many coastal wetlands has occurred over the previous two centuries in order to accommodate agriculture, aquaculture and urban development (Armentano & Menges, 1986; Villa & Bernal, 2018; White et al., 1997). Drained wetlands can lead to rapid soil organic matter oxidation, and transform systems to net CO$_2$ sources (Deverel et al., 2016; Pereyra & Mitsch, 2018). Drainage systems can also reduce wetland inundation...
periods and alter sediment redox-dependant geochemistry and microbially-mediated reactions (Johnston et al., 2014), particularly those involving bioavailable iron (Fe(III)), sulphate (SO$_4^{2-}$) and nitrate (NO$_3^-$). Importantly, anaerobic carbon metabolism employing these terminal electron acceptors (Fe(III), SO$_4^{2-}$, NO$_3^-$) competes thermodynamically with methanogenic bacteria and archaea and thereby can inhibit CH$_4$ production (á Norði & Thamdrup, 2014; Burdige, 2012; Karimian et al., 2018; Lal, 2008). With increasing value now placed on the ecosystems services provided by wetlands, many degraded systems are now undergoing remediation and re-flooding (Johnston et al., 2014). However, the ecosystem benefits, such as enhanced biodiversity and water quality, may come at a price in the form of high initial CH$_4$ flux rates, and predicted net radiative forcing for several centuries post-remediation - thus posing a ‘biogeochemical compromise’ (Hemes et al., 2018; Lal, 2008).

Within Australia, it has been estimated that more than 50% of natural wetlands have been lost to land use change, drainage and degradation since European settlement (Finlayson & Rea, 1999; ANCA, 1995). This equates to an estimated ~1.2 Pg C emitted to the atmosphere through oxidation of soil organic carbon (Page & Dalal, 2011). Much of eastern Australia’s freshwater coastal wetlands are underlain by Holocene derived sulphidic sediments (i.e pyrite – Fe$_2$S, known as coastal acid sulphate soils; CASS) formed during periods of higher sea levels (Walker, 1972; White et al., 1997). When CASS are drained, pyrite is oxidised, producing sulphuric acid (H$_2$SO$_4$). This results in highly acidic soils with pH levels as low as 3 (Johnston et al., 2014; Sammut et al., 1996). After rainfall events, groundwater transports H$_2$SO$_4$ from the CASS landscapes into nearby creeks and estuaries (Sammut et al., 1996). The low pH groundwater discharge also mobilises iron and aluminium, fuels aquatic deoxygenation, and can lead to large fish kills and degradation of infrastructure (Jeffrey et al., 2016; Johnston et al., 2003; White et al., 1997; Wong et al., 2010). Drained CASS wetlands typically contain abundant reactive Fe(III) and exhibit complex sulphur and Fe cycling (Boman et al., 2008; Burton et al., 2011; Burton et al., 2006). Wetland iron and sulfur cycling can profoundly influence CH$_4$ production and consumption via a series of complex redox reactions coupled with organic matter mineralisation (Holmkvist et al., 2011; Sivan et al., 2014). As such, terminal electron acceptor availability is critical when considering wetland remediation and the biogeochemical compromise paradigm.

Here we assess CH$_4$ emissions rates from a remediated freshwater CASS wetland in subtropical eastern Australia, and compare fluxes from the permanent wetland and the adjacent seasonal wetland ecotypes. We hypothesize that wetland CH$_4$ emissions will differ
significantly between the seasons and between the four wetland communities. We account for three atmospheric flux pathways for methane; ebullition, diffusion and plant-mediated fluxes, over diurnal cycles and within different seasons. CH₄ fluxes were also assessed in relation to the underlying soil properties, including sulphate, reactive iron III and iron II, acid volatile sulphur, chloride and organic carbon.

2.0 Methods

2.1 Study site

Cattai Wetland is located on the mid-coast of New South Wales, Australia. The reserve covers 500 hectares, featuring a shallow permanent wetland covering an area of approximately 16 hectares that is adjacent to a seasonal wetland and floodplain located to the south (Fig. 1). Both sites discharge into the nearby Coopernook Creek, a tributary of the larger Manning River estuary. The site was extensively cleared and low-lying areas drained during the early 1900’s in order to aid agriculture and development in the region. As a result of this anthropogenic drainage, the oxidation of CASS produced sulphuric acid and episodic acidic discharge to adjacent creeks for many years (Tulau, 1999). To ameliorate acidic discharge, the natural hydrology of the site was restored in 2003 through the decommissioning of agricultural drains and removal of floodgates. Re-flooding of the CASS landscape has reduced the production of sulphuric acid, acid discharge and aluminium and iron mobilisation, hence improving the downstream water quality (GTCC, 2014).

The region receives a mean annual rainfall of 1180 mm with the majority falling during early autumn with an average maximal monthly rainfall in March (152 mm). The lowest rainfall generally occurs during the winter months with average minimal rainfall during September (60 mm). Average minimum and maximum summer temperatures range from 17.6 °C to 29 °C (January) and in winter range from 5.9 °C to 18.5 °C (July) (BOM, 2018). The dominant vegetation type within the permanent wetland is an introduced waterlily species (Nymphaea capensis), while the fringes of the wetland consist of wetland tree species; Casuarina sp. and Melaleuca quinquenervia. The seasonal wetland to the south is dominated by the sedge; Juncus kraussii (Veg A) and features scattered stands of Phragmites australis (Veg B) with areas of slightly higher elevation dominated by Juncus kraussii below Casuarina sp. (Veg C) (Fig. 1).
2.2 The aquatic CH$_4$ flux of the permanent wetland

To quantify CH$_4$ ebullition rates, up to 12 ebullition domes were deployed during two distinct seasons (detailed below) at ~20 m intervals along a longitudinal transect, from the edge of the permanent wetland towards the centre. Each dome was carefully suspended below the water level by flotation rings, ensuring minimal disturbance of sediment and the water column. Gas samples were extracted from the headspace of each dome using a 300 mL gas tight syringe at periods of ~48 h. The volume was recorded and each sample then diluted using ambient air (1:729 ratio) and analysed in situ using a manufacturer calibrated cavity ring-down spectrometer (Picarro G2201-i) to determine CH$_4$ concentrations (ppm). Diffusive CH$_4$ fluxes from the permanent wetland were measured using a floating chamber with a portable greenhouse gas analyser (UGGA, Los Gatos Research). To account for spatial and temporal variability, measurements were conducted during both day-time and night-time, and sampling within vegetated areas featuring lilies (*Nymphaea capensis*), forested areas (*Melaleuca* sp.) and in areas where no aquatic vegetation was present (i.e. open water). A total of 39 CH$_4$ floating chamber incubations averaging ~8 minutes in duration were recorded over the two campaigns. The average $r^2$ value of linear regressions of CH$_4$ concentrations versus time during chamber incubations was 0.97 ± 0.05. One chamber measurement was disregarded as an outlier (as it was more than three times the standard deviation of the mean) and any chambers capturing ebullition bubbles (determined by a nonlinear increase in concentration) were also disregarded. The seasonal ebullition and diffusive CH$_4$ flux methods and measurements from the permanent wetland have previously been reported elsewhere (Jeffrey et al. submitted).

2.3 Plant-mediated CH$_4$ fluxes

Simultaneous time series chamber experiments were conducted over ~24 hours to measure CH$_4$ fluxes during each season from the three different wetland vegetation ecotypes. These ecotypes were *Juncus kraussii* (Veg A), *Phragmites australis* (Veg B) and *Juncus kraussii* amongst *Casuarina* sp. forest (Veg C) (Fig. 1). In each ecotype, 65 x 65 x 30 cm acrylic bases were installed four months before the first time series experiment, to minimise disturbance to the sediment profile and vegetative rhizosphere. Vegetative flux chambers were constructed of an aluminium frame with clear Perspex walls and roof that matched the areal footprint of the pre-inserted acrylic bases. The chambers were 100 cm, 150 cm and 50 cm high for at Veg A, B and C respectively. The custom sizes were tailored for the different vegetation
heights, whilst minimising chamber volume as much as possible. Each chamber was leak-tested under laboratory conditions prior to fieldwork.

Before each field incubation, chambers were flushed with atmospheric air then carefully lowered over the vegetation and onto the acrylic base ensuring an air tight seal. A small fan circulated internal air within each chamber. Air within the chamber was pumped through a closed loop from the top of the chamber using gas tubing (Bevaline), passing through a drying agent (Drierite desiccant) and then analysed in situ using a calibrated cavity ring-down spectrometers (Picarro G2201-i or LosGatos), recording the flux rate of CH₄ (ppm/sec). The gas flow was returned near the base inside each vegetation chamber closing the loop. Vegetation incubation times ranged from 6 to 15 minutes depending on the flux rate and were taken from triplicate sites to account for heterogeneity within each ecotype. The daytime measurements (after sunrise) were measured at ~10 minute intervals whilst night time measurements (after sunset) were taken at ~4 hourly intervals. CH₄ fluxes from the adjacent exposed sediments or shallow overlying water at each site were also measured at ~4 hourly intervals to determine the influence and role of plant-mediated CH₄ fluxes compared to non-vegetated CH₄ fluxes. Light and temperature loggers (Onset Hobo) measured the changes in diurnal air temperature (°C) and photosynthetically active radiation (PAR) at each site.

2.4 Soil geochemistry and redox conditions

A water logger (Minidiver) was deployed in the permanent wetland before the first campaign to monitor changes in water depth (cm) and temperature (°C). Field pH (pH₅) and the redox potential (EhF: reported against standard hydrogen electrode) were determined in situ, by directly inserting the electrode into the soils (5 cm depth, 8 replicates) at each site. A composite sampling approach (3 cores) was used to collect sediment samples from each site, to determine organic C content, Fe(III)HCl, Fe(II)HCl, Cl, SO₄²⁻ and acid volatile sulphur (AVS). The cores were extracted in December 2016, by inserting a 4.0 cm diameter acrylic tube into the sediment to a depth of up to 50 cm. Cores were immediately sectioned into 2 cm increments to a depth of 20 cm, and 5 cm increments thereafter, ensuring higher vertical resolution in the organic rich near-surface sediments. Samples were immediately placed into air-tight bags, then frozen within 12 hr of collection at -16°C in a portable freezer and transferred to -80°C freezer in the laboratory. Frozen samples were thawed in an oxygen-free anaerobic chamber (1-5% H₂
in N\textsubscript{2}), using an oxygen consuming palladium (Pd) catalyst. The defrosted samples were homogenised using a plastic spatula.

AVS content was determined by adding 1-2 g of wet sediment with 6 M HCl:1 M L-ascorbic acid. The liberated H\textsubscript{2}S was captured in 5 ml of 3% Zn acetate in 2 M NaOH and then quantified using iodometric titration. The reactive Fe fractions were determined using a sequential extraction procedure optimised for acid sulphate soils based on Claff et al. (2010). Poorly crystalline solid-phase Fe (II) and Fe (III) were determined by extracting 2 g wet sub-samples with cold N\textsubscript{2}-purged 1 M HCl for four hours. Aliquots of 0.45 µm-filtered extract were analysed for Fe (II) [Fe(II)\textsubscript{HCl}] and total Fe [Fe\textsubscript{HCl}] using the 1,10-phenanthroline method with the addition of hydroxylammonium chloride for total Fe (APHA, 2005). The Fe(III) [Fe(III)\textsubscript{HCl}] was determined by the difference of [Fe\textsubscript{HCl}] – [Fe(II)\textsubscript{HCl}]. Total organic carbon (TOC) and total S (S\textsubscript{Tot}) were determined via a LECO CNS-2000 carbon and sulfur analyser. Chloride and sulfate concentrations were measured using filtered (0.45 µm) aliquot from a 1:5 water extract of freshly defrosted wet soil, as per Rayment and Higginson (1992) via ion chromatography using a Metrosep A Supp4-250 column, an RP2 guard column and eluent containing 2 mM NaHCO\textsubscript{3}, 2.4 mM Na\textsubscript{2}CO\textsubscript{3} and 5% acetone, in conjunction with a Metrohm MSM module for background suppression.

2.5 Calculations

Both the air-water and vegetative CH\textsubscript{4} fluxes were calculated for the chamber deployments in the permanent wetland and seasonal wetland using the equation:

\[ F = (s(V/RT_{air}A))t \]  \hspace{1cm} (1)

where \( s \) is the regression slope for each chamber incubation deployments (ppm sec\textsuperscript{-1}), \( V \) is the chamber volume (m\textsuperscript{3}), \( R \) is the universal gas constant, \( T_{air} \) is the air temperature inside the chamber (K), \( A \) is the surface area of the chamber (m\textsuperscript{2}) and \( t \) is the conversion factor from seconds to day, and to mmol.

Ebullition rates (\( E_b \)) (mmol m\textsuperscript{-2} d\textsuperscript{-1}) were calculated using the equation:

\[ E_b = ([CH_4].CH_4Vol.)/ A.V_m.Td \]  \hspace{1cm} (2)
where \([\text{CH}_4]\) is the CH\(_4\) concentration in the collected gas (%), \(\text{CH}_4\text{Vol.}\) is the gas volume sampled (L), \(A\) is the funnel area (m\(^2\)), \(V_m\) is the molar volume of CH\(_4\) at in situ temperature (L) and \(T_d\) is deployment time (days).

### 3.0 Results

#### 3.1 Hydrological Conditions

Prior to the first campaign in April 2017 (C1), an extreme hot/drying summer period occurred during early 2017 (Fig. 2). This resulted in an average wetland water column temperature of 23.3 ± 0.7 °C and a water depth in the permanent wetland as low as ~7.3 cm, with exposed sediments along the wetland perimeter during the preceding month. Total rainfall for the two weeks prior to C1 was 342 mm, with an additional 35 mm of rain occurring during C1 fieldwork (Fig. 2) thus raising the water column depth in the permanent wetland to 77.2 cm in less than four weeks. This C1 deployment was therefore categorized as the ‘post-dry/flooded’ period, where air temperatures ranged from 13.3 to 22.8 °C and the average water column temperature in the permanent wetland was 20.4 ± 0.5 °C. The second fieldwork campaign was conducted in September 2017 (C2) under cool/drying conditions, where air temperatures ranged from as low as 3.4 °C to 34.9 °C (Fig. 2), with cooler average water temperatures 12.6 ± 0.4 °C in the permanent wetland (Fig. 2). The depth of the permanent wetland at this time had dropped slightly to ~33 cm (Fig. 2).

#### 3.2 Permanent and Seasonal Wetland CH\(_4\) fluxes

The vegetation time series revealed diurnal variability of plant-mediated CH\(_4\) emissions occurred at most ecotypes, with the highest CH\(_4\) fluxes occurring during daytime around midday and the lowest CH\(_4\) fluxes during the night time (Fig. 3, Table 1). The lowest CH\(_4\) fluxes were found at Veg C with a net negative CH\(_4\) flux observed during C2 time series. The CH\(_4\) sediment fluxes measured amongst each vegetation time series were consistently much lower than the plant-mediated CH\(_4\) fluxes indicating that the vegetation was indeed the main conduit for CH\(_4\) to the atmosphere (Fig. 3, Table 1). The CH\(_4\) fluxes were highly variable between the replicates at each site. Temperature and PAR followed similar diurnal trends to each other and had positive correlations to CH\(_4\) emissions (Fig. 3).
CH$_4$ fluxes from the three vegetation types were higher in C1 than C2 (Fig. 4, Table 1). The highest CH$_4$ fluxes in each of the vegetation types always occurred during the daytime (Fig. 4, Table 1). Phragmites sp. (Veg B) consistently emitted the highest CH$_4$ fluxes (2.27 ± 1.42 mmol m$^{-2}$ d$^{-1}$ during C1 and 0.77 ± 0.46 mmol m$^{-2}$ d$^{-1}$ during C2). The Veg C ecotype within the seasonal wetland consistently produced the lowest CH$_4$ fluxes of all sites, with a net negative flux occurring during C2 (-0.01 ± 0.08 mmol m$^{-2}$ d$^{-1}$).

The permanent wetland showed an inverse trend with seven-fold higher diffusive fluxes during the cool/drying C2 (10.46 ± 15.81 mmol m$^{-2}$ d$^{-1}$) compared to the post-dry/flooded C1 (1.49 ± 2.75 mmol m$^{-2}$ d$^{-1}$), while the ebullition rates were similar during both seasons (Fig. 4, Table 1). Overall, the plant mediated CH$_4$ fluxes from the three seasonal wetland vegetation ecotypes (Veg A, B and C) were within the range of aquatic fluxes measured from the permanent wetland for the post-dry/flooded C1 time series, but not for the cool/drying C2 time series, when the permanent wetland CH$_4$ fluxes were much higher (Fig. 4).

### 3.3 Sediment core profiles and soil redox potentials

Average concentrations from soil cores (Table 1, Fig. 5) were based upon the top 20 cm of the profile, where the highest organic carbon concentrations were found. This upper rhizosphere depth zone is assumed to be an active area of carbon metabolism and CH$_4$ production and consumption (Nedwell & Watson, 1995). The Fe(III)$_{HCl}$ concentrations were greater than Fe(II)$_{HCl}$ at all three seasonal wetland sites, however the permanent wetland showed an opposite trend with low concentrations of both Fe(III) (5.6 ± 10.7 mmol kg$^{-1}$) and SO$_4^{2-}$ (1.5 ± 1.0 mmol kg$^{-1}$) (Fig. 5, Table 1). The highest average concentrations of Fe(III)$_{HCl}$ were found at the Veg C site (204.0 mmol kg$^{-1}$) and highest and similar concentrations of SO$_4^{2-}$ were in Veg B and Veg C sediments (45.4 ± 41.0 mmol kg$^{-1}$ and 43.3 ± 16.7 mmol kg$^{-1}$) (Fig. 5, Table 1). Net positive redox potential was found at all four sites during C1 (under post-dry/flooded conditions) indicating a lag time between recent flooding and the onset of reducing conditions. In contrast, a negative redox potential was found within the permanent wetland and Veg B during C2, indicating reduced conditions under cool drying conditions (Table 1). The TOC concentrations (%) were highest in the upper profiles and similar across all sites (Fig. 5, Table 1) averaging 13.4 ± 7.6%.
3.4 Temperature and PAR

Correlation plots for both temperature (ºC) and sunlight (PAR) versus CH4 emissions from the three vegetation ecotypes showed no distinct relationships with the exception of Veg B during C2 for PAR ($r^2=0.18$, $p<0.01$) and temperature (ºC) ($r^2=0.35$, $p<0.001$). No clearer trends were observed by combining all site measurements, nor separating daytime fluxes and drivers from night time fluxes and drivers.

4.0 Discussion

4.1 Geochemistry of the CASS landscape

Sediment profiles provide insights to the historical geochemical changes that have occurred across the CASS landscapes of the four Cattai Wetland sites (Fig. 5). If we assume that relatively uniform deposition of late Holocene materials occurred, the differences between present day profiles are related to historical changes in hydrology and land use, topographic elevation, geochemical trajectories and vegetative carbon inputs. For example, the permanent wetland shows distinct differences to the adjacent seasonal wetland sites, with divergent geochemical signatures of both iron and sulphate that reflect the sustained inundation (Table 1, Fig 5). The permanent wetland had significantly lower Fe(III) ($p<0.001$) and 11 to 30 fold lower SO$_4^{2-}$ concentrations within the upper soil profile compared to the seasonal wetland. The ratio of Fe(III)$_{HCl}$ to Fe(II)$_{HCl}$ from the flooded soils of the permanent wetland was 0.03, indicating the sediments were almost completely depleted of Fe(III). Under reducing conditions where there is low SO$_4^{2-}$ and little to no Fe(III) to competitively exclude methanogenesis, CH$_4$ production becomes more favourable. Indeed, CH$_4$ production was on average highest from the permanent wetland, especially when considering the duel CH$_4$ pathways of ebullition and air-water diffusion (Table 1).

In addition to sulphate reduction, some depletion of the sulphur pool from the permanent wetland may have occurred due to drainage exports of sulphuric acid (H$_2$SO$_4$) discharging from the CASS landscape throughout the last century. Alternatively, reducing conditions induced by re-flooding freshwater wetlands is known to encourage the re-formation of AVS and pyrite (FeS$_2$) and produce alkalinity, thereby attenuating acid production and discharge (Burton et al., 2007; Johnston et al., 2014; Johnston et al., 2012) and reducing the total SO$_4^{2-}$ pool of CASS landscapes. While the AVS concentrations found within the
permanent wetland (up to 18.5 µmol g\(^{-1}\)) were a result of sulphate reduction induced by CASS
wetland restoration, they nonetheless represent a relatively volatile form of sulphur, which is
at risk of rapid oxidation during drought periods (Johnston et al., 2014; Karimian et al., 2017).
The AVS concentrations of the permanent wetland sites were more than 20-fold higher than
the three adjacent seasonal wetland sites, and represent a potentially volatile by-product and
consequence of re-flooding CASS soil landscapes, in addition to leading to increases of CH\(_4\)
emissions (Table 1).

The soil profile from the seasonal wetland Veg C habitat featured abundant Fe(III)\(_{\text{HCl}}\)
(Fe(III)\(_{\text{HCl}}\) to Fe(II)\(_{\text{HCl}}\) ratio of 136) and also SO\(_4^{2-}\). This was associated with the lowest fluxes
of CH\(_4\) for both seasonal sampling periods (Fig. 5, Table 1). Relatively low CH\(_4\) fluxes from
Veg C are likely due to the more oxidising conditions present at this site and the surfeit of
thermodynamically favourable terminal electron acceptors (i.e. Fe(III) and SO\(_4^{2-}\)), which
would competitively exclude organic matter degradation by methanogenic archaea (Postma &
Jakobsen, 1996).

At the other seasonal wetland sites (Veg A and B), the average Fe(III) and SO\(_4^{2-}\)
concentrations were intermediate, (i.e. lower than Veg C, but higher than the permanent
wetland), although in the upper profile Veg B had more SO\(_4^{2-}\) while Veg A had more Fe(III)
(Fig. 5, Table 1). CH\(_4\) flux values from these sites were also intermediate (Table 1). Sediment
profiles from both Veg A and Veg B indicated a degree of Fe reduction based on the ratio of
Fe(III):Fe(II) which were 7.2 and 3.6 respectively. The redox potentials from Veg B during
both C1 and C2 seasons (9.6 mV and -89.0 mV respectively) were consistently lower than Veg
A during C1 and C2 seasons (46.5 mV and 12.0 mV respectively), which is consistent with the
more reducing conditions encouraging CH\(_4\) production in Veg B habitat. Further, as iron
reduction yields more free energy than SO\(_4^{2-}\) reduction (which yields more free energy than
methanogenesis) (Burdige, 2012), then Fe reduction at Veg A may outcompete CH\(_4\) production
ahead of SO\(_4^{2-}\) reduction at Veg B, which may help explain some of the differences in CH\(_4\)
production between the two sites.

Regression analysis and Spearman rho coefficients summarise the spatial trends
occurring between the average sediment parameters versus seasonal CH\(_4\) fluxes from the
different sites (Fig. 7). Positive significant trends occurred for Fe(II), AVS and the Cl:SO\(_4^{2-}\)
ratios with CH\(_4\) flux rates (\(r_s=0.88, p<0.01\)) supporting our hypothesis that reducing conditions
and a smaller pool of sediment Fe(III) and SO\(_4^{2-}\) facilitate higher CH\(_4\) production rates.
Alternatively, the negative trends observed between soil redox potentials, SO$_4^{2-}$, Fe(III) and CH$_4$ fluxes affirm that the abundance of thermodynamically favourable terminal electron acceptors plays a role in attenuating CH$_4$ production at each site.

4.2 Plant-mediated CH$_4$ fluxes from the seasonal wetland

Plant-mediated CH$_4$ fluxes were highest during C1 under post-dry/flooded conditions with 20-30 cm of standing waters in the seasonal wetland (Table 1). While waterlogged conditions are an obvious driver of higher CH$_4$ production rates from saturated sediments in addition to the geochemical differences (previously discussed), other drivers which may explain these trends include differences in diurnal variability in temperature, PAR and plant physiology, which may influence CH$_4$ gas transport pathways.

In vegetated seasonal wetlands, plant-mediated gas transport is recognised as a dominant pathway for CH$_4$ emission to the atmosphere and accounts for up to 90% of total wetland fluxes (Sorrell & Boon, 1994; Whiting & Chanton, 1992). For plant survival in near-permanent inundation environments, oxygen transport occurs via the aerenchyma downwards to the rhizome. This increases the plant performance by mitigating (i.e. oxidising) the accumulation of phytotoxins such as sulphides and reducing metal ions around the roots (Armstrong & Armstrong, 1990; Armstrong et al., 2006; Penhale & Wetzel, 1983). As oxygen transfer to the rhizosphere occurs, an exchange of sedimentary CH$_4$ can be efficiently transported from the rhizosphere to atmosphere, bypassing sedimentary oxidative processes along the way (Fig. 8). This process in plants can be either convective (i.e. pressurised) or via passive diffusive gas flow, both of which are adaptive traits of many wetland species (Armstrong & Armstrong, 1991; Konnerup et al., 2011).

During both seasons the highest CH$_4$ fluxes from seasonal wetland vegetation were emitted from *Phragmites australis* (Veg B) and always occurred during daylight (Table 1, Fig. 8). In *Phragmites australis* (Veg B), the presence of pressurised lacunar leaf culms drive a mass flow of oxygen to the rhizome and back to the atmosphere via older (non-pressurised) efflux culms (Henneberg et al., 2012; Sorrell & Boon, 1994). This process has been widely studied in wetlands featuring this species, as it is one of the most productive and wide spread flowering wetland species (Brix et al., 2001; Chanton et al., 2002; Clevering & Lissner, 1999; Tucker, 1990). Kim et al. (1998) showed CH$_4$ emissions from *Phragmites australis* peaked around midday and that daytime emissions were about 3-fold higher than night time emissions,
positively correlating with temperature and PAR. These were similar to our findings with highest CH$_4$ fluxes of each seasonal time series occurring near midday (10:50 am during C1; 4.88 mmol m$^{-2}$ d$^{-2}$ and 12:15 pm during C2; 2.06 mmol m$^{-2}$ d$^{-2}$) (Fig. 3). We also found a positive significant relationship between CH$_4$ flux and both temperature and PAR during C2 ($r^2=0.35$, p<0.001 and $r^2=0.18$, p<0.01 respectively) (Fig. 6). The often high diurnal variability in CH$_4$ fluxes from *Phragmites australis* occurs as convective gas transport increases rhizospheric oxygen and CH$_4$ exchange via living culms during the daytime, whereas molecular diffusion during the night time facilitates a more passive and lower CH$_4$ flux pathway through dead culms (Armstrong & Armstrong, 1991; Chanton et al., 2002).

One possible reason CH$_4$ fluxes were lower from Veg A than Veg B despite their close geographical location, may be due to the passive gas diffusion mechanism utilised by *Juncus sp.* (Henneberg et al., 2012). Unlike the pressurised conductive gas flow mechanisms of Veg B, many wetland rush species (such as Veg A) employ passive diffusive gas flow to survive within water logging environments (Brix et al., 1992; Konnerup et al., 2011). Despite diffusion being a less efficient gas transport mechanism (Konnerup et al., 2011), plant-mediated CH$_4$ diffusion is recognised as the dominant pathway for CH$_4$ emissions from many seasonal wetland species. During C1 and C2, day time fluxes (diffusive) from Veg A were only 19% and 33% higher than night time fluxes (diffusive). In comparison, at Veg B these day:night ratios were almost triple this (67% and 94% higher) during the same periods. This may potentially be due to the more efficient daytime conductive gas transfer pathway of CH$_4$ through Veg B (*Phragmites australis*) compared to the more passive diffusive CH$_4$ gas transfer pathway of Veg A (*Juncus kraussii*). This suggests that non-pressurized pathways may result in lower net rhizosphere-atmosphere gas exchange of CH$_4$ from seasonal wetland vegetation.

The *Juncus kraussii* below *Casuarina sp.* trees (Veg C) emitted nominal fluxes of CH$_4$ during both time series campaigns and was a net sink for CH$_4$ during C2 (Table 1, Fig. 8). Although wetland trees have recently been shown to contribute significantly to CH$_4$ fluxes from flooded environments (Pangala et al., 2017), we could not quantify or constrain the role of trees as a conduit of methane to the atmosphere at this site. Regardless, there were clearly lower CH$_4$ fluxes through the Veg C (*Juncus kraussii*) compared to the Veg A (*Juncus kraussii*). As the species at ground level were identical, these differences are not related to vegetative gas transport mechanisms, nor organic carbon content (Table 1). Shading by the overhanging trees may inhibit the daytime diffusive CH$_4$ gas transport through Veg C assumable to lower rates of photosynthesis, however PAR was only lower during C2 (Fig. 7).
and so does not appear to explain the CH₄ flux differences observed during C1. The differences are therefore likely explained by the higher positive redox potentials (Table 1) and more abundant thermodynamically favourable terminal electron acceptors (i.e. Fe(III) and SO₄²⁻) (Fig. 5) all of which can inhibit methane production within the sediments (Burdige, 2012).

4.3 Permanent Wetland CH₄ fluxes

Diffusive CH₄ fluxes from the permanent wetland varied considerably between seasons; however, ebullition fluxes were similar (Table 1, Fig. 8). The highest seasonal CH₄ fluxes for both ebullition and diffusion (2.1 mmol m⁻² d⁻¹ and 10.5 mmol m⁻² d⁻¹ respectively) occurred during C2 despite cooler conditions (Fig. 2. Fig. 8). This however was the opposite trend to the seasonal wetland CH₄ fluxes (Table 1, Fig. 8). One reason may be due to the antecedent hydrological conditions before C1 (Fig. 2). Jeffrey et al (submitted) reported that a water level drawdown of the permanent wetland after a hot and drying summer period exposed some of the permanent wetland sediments to oxidative conditions. This may have oxidised a portion of the labile sedimentary carbon pool prior to C1 sampling of the permanent wetland, therefore reducing the total CH₄ pool observed during C1 sampling. A lag time (ranging from weeks to months) for recovery of the CH₄ pool post-drought has been observed in other systems (Boon et al., 1997) and also during lab-based experiments (Freeman et al., 1992; Knorr et al., 2008). This may explain the higher CH₄ fluxes during C2 when the system had had sufficient time to recover, despite lower water column temperatures that would normally reduce microbial metabolism rates. This hypothesis is also supported by the shift of net positive redox potential of the permanent wetland during C1 (71.7 ± 65 mV), to a strong negative redox potential during C2 (-216 ± 42 mV) indicating that there was a time lag for reducing conditions to recover within the permanent wetland for C2. This highlights the critical role of antecedent hydrological conditions and how dynamic weather oscillations of drought and floods (a common occurrence of many Australian wetland systems), strongly influence the redox potentials, soil geochemistry and ultimately CH₄ fluxes.

4.4 Implications and conclusions

Permanent wetland emissions account for the majority of the global wetland CH₄ budget however both subtropical systems and southern hemisphere systems are poorly
represented (Bartlett & Harriss, 1993; Bastviken et al., 2011) (Fig. 9). Further, the fluxes from seasonal wetlands are poorly constrained (Pfeifer-Meister et al., 2018) due to their intermittent nature and variability of intra-seasonal areal extent, which may compound why natural wetlands have the largest uncertainty of the global methane budget (Kirschke et al., 2013; Saunois et al., 2016). Although the temporal resolution of our study cannot be up scaled to realistic annual estimates, our high resolution sampling strategy provided insights to daily CH$_4$ flux rates revealing distinct differences between different vegetation types across the terrestrial aquatic wetland boundary. Our seasonal emissions rates were at the low end of the scale of measurements made in southern hemisphere subtropical systems but within range of northern hemisphere subtropical systems of similar latitudes (Fig. 9).

Although remediating degraded wetlands through re-flooding is a common technique to improve biodiversity, increase C sequestration and improve downstream water quality issues (Johnston et al., 2014; Johnston et al., 2004), our results propose a nuanced dilemma for land use managers, as wetland restoration can have net positive radiative forcing effects on the Earth’s climate due to high rates of CH$_4$ production (Mitsch et al., 2013). This has also been shown to be particularly high during early remediation periods (Hemes et al., 2018). Our results suggest that seasonal wetlands emit less CH$_4$ on an areal basis than permanent wetlands, yet carbon accumulation in these soils may be lower (Brown et al. (in publication)). Longer-term studies over annual cycles encompassing seasonal drivers and CH$_4$ fluxes would further test this hypothesis of the different drivers between seasonal and permanent wetland systems.

Our results also suggest that selective hydrological restoration of wetlands featuring sediments with abundant thermodynamically favourable terminal electron acceptors (i.e. Fe(III) or SO$_4^{2-}$) may be a (partial) biogeochemical solution (also suggested by Hemes et al. (2018)) to both remediate degraded sites whilst simultaneously mitigating some CH$_4$ emissions. When Fe(III) and SO$_4^{2-}$ are abundant in anaerobic environments they provide preferential terminal electron acceptors for microbial metabolism and thus limit methanogenesis via competitive exclusion (Achtnich et al., 1995). However, high rates of sulphate reduction coupled with Fe reduction can also lead to the accumulation of metal sulphide minerals e.g. pyrite and AVS (Johnston et al., 2014). Under permanently saturated and low oxygen conditions, metal sulphides will steadily accumulate and remain relatively benign. However, if the saturated state of remediated sites cannot be maintained, AVS may react with oxygen resulting in undesirable production of acidity and low pH conditions. Therefore the remediation of wetlands for carbon storage should involve careful site selection.
to both limit CH₄ production and to avoid redox related geochemical by-products with detrimental environmental effects.

This study has highlighted how sediment geochemistry is intimately related to CH₄ production and consumption. While high sulphate and Fe(III) favour lower CH₄ production, sites featuring more reducing conditions and depleted sulphate and Fe(III) favour the highest CH₄ fluxes. Results reveal distinct differences between the areal CH₄ fluxes of four different eco-types located within a remediated subtropical Australian wetland and indicate high seasonal variability. By combining novel and well established techniques we delineated several CH₄ pathways of both seasonal and permanent wetland sources (ebullition, diffusion and plant-mediated pathways) and linked these to seasonal drivers. This provided evidence that soil geochemistry is an important factor to consider for wetland remediation in the context of CH₄ production and mitigation strategies. The CH₄ emissions results were comparable to other wetlands of similar latitudes and contribute important data for both the understudied southern hemisphere wetlands and seasonal subtropical wetland ecotypes.

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Fig. 1 The seasonal wetland study sites consisting of Veg A (*Juncus kraussii*), Veg B (*Phragmites australis*), Veg C (*Juncus kraussii* below *Casuarina sp.*) and the permanent wetland indicating sediment coring sites, ebullition replicate transect, 24 h vegetation time series sites and imagery of vegetation ecotypes.
Figure 2. Hydrograph for the seven months of 2017 indicating daily rainfall, maximum/minimum air temperature, water temperature and antecedent hydrology. Vertical coloured bands represent the two fieldwork campaigns.
Figure 3. Simultaneous 24 h time series of vegetative CH₄ fluxes from the seasonal wetland ecotypes at Cattai Wetland during C1: post-dry/flooded (Apr 2017) and C2: cool/drying conditions (Sep 2017). The vertical error bars of the plant-mediated CH₄ flux (mmol m⁻² d⁻¹) represent standard deviation of the triplicate time series measurements taken from each site and horizontal bars represent the total aggregated time period represented by replicate chambers. The grey shading indicates night-time. Note: Different y-axis scales for CH₄ to highlight diurnal trends.
Figure 4. Seasonal fluxes of CH$_4$ from diurnal sampling and ebullition from the permanent wetland and adjacent 24 h time series of the seasonal wetland vegetation types A, B and C. Note: Dashed line represents the average, solid line represents the median and dots represent 5$^{th}$ and 95$^{th}$ percentiles.
Figure 5. Soil profiles of the permanent and seasonal wetland sites indicating Fe(II)$_{\text{HCl}}$, Fe(III)$_{\text{HCl}}$, SO$_4^{2-}$, Cl:SO$_4^{2-}$ (a proxy for depletion of marine-derived sulphate, where >20 is broadly indicative of SO$_4^{2-}$ reduction and <8 CASS pyrite oxidation (Mulvey, 1993)), total C and acid volatile sulphur (AVS). Note: The permanent wetland profiles are averages from two adjacent sites with error bars representing the standard deviation.
Figure 6. Correlations of CH$_4$ with temperature (ºC) and photo-synthetically active radiation (PAR) (lum ft$^{-2}$) for the three seasonal wetland vegetation sites of Cattai Wetland during two seasonal campaigns.
Figure 7. Regression analysis of average daily CH$_4$ fluxes (mmol m$^{-2}$ d$^{-1}$) vs subsoil parameters of 0-20 cm core depth (i.e. CH$_4$ ‘active’ zone). Note: Log scale y-axis of CH$_4$ fluxes from the four wetland ecotypes over two seasons. Note: The $r_s$ values calculated using Spearman rho are for C1 (black shapes) and C2 (white shapes).
Figure 8. Conceptual model summarising the terrestrial and aquatic CH$_4$ fluxes (mmol m$^{-2}$ d$^{-1}$) and sediment core profile parameters (mmol kg$^{-1}$) of the permanent and seasonal wetlands during C1 (post-dry/flooded conditions) and C2 (cool/drying conditions) of Cattai Wetland. Conceptual diagram expanded from Jeffrey et al. (in publication) and rhizome process insert adapted from (Conrad, 1993).
Figure 9. Summary of major CH₄ wetland reviews by Bartlett and Harriss (1993), Bastviken et al. (2011) and modelled fluxes by Cao et al. (1998) adapted from Jeffrey et al., (in publication) highlighting latitudinal trends and bias from a variety of wetland systems. Inset figure highlights number of studies in these reviews by latitudinal increments of 10° poleward of the equator. Note: x axis scaled to highlight subtle differences between studies.
**List of Tables**

**Table 1.** Summary of plant-mediated CH$_4$ fluxes from the seasonal wetland time series and diurnal CH$_4$ fluxes and ebullition from the permanent wetland during C1 (post-dry/ flooded) and C2 (cool/ drying). The corresponding sediment core data are average concentrations from 0 to 20 cm below ground level.

<table>
<thead>
<tr>
<th>CH$_4$ flux (mmol m$^{-2}$ d$^{-1}$)</th>
<th>Ebulition</th>
<th>Diffusion</th>
<th>Veg A</th>
<th>Veg B</th>
<th>Veg C</th>
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<tbody>
<tr>
<td>C1 - Sediment flux</td>
<td></td>
<td></td>
<td>0.06</td>
<td>0.04</td>
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<tr>
<td>C1 - Day time</td>
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<td>1.79</td>
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<tr>
<td>C1 - Daily average</td>
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<td>1.49</td>
<td>1.70</td>
<td>2.27</td>
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<td>C2 - Sediment flux</td>
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<td></td>
<td>0.0004</td>
<td>0.20</td>
<td>0.0003</td>
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<tr>
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<td>0.94</td>
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<td>0.04</td>
<td>0.48</td>
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<td>10.46</td>
<td>0.05</td>
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Sediment core average (0-20cm)

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<th>Seasonal Wetland Sites</th>
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<tbody>
<tr>
<td>Fe$_{HCl}$ (II) (mmol kg$^{-1}$)</td>
<td>202.3</td>
<td>11.6</td>
</tr>
<tr>
<td>Fe$_{HCl}$ (III) (mmol kg$^{-1}$)</td>
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<td>SO$_4^{2-}$ (mmol kg$^{-1}$)</td>
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<td>ClSO$_4^{2-}$</td>
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<td>8.4</td>
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<tr>
<td>AVS (µmol g$^{-1}$)</td>
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<td>0.7</td>
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<td>TOC (%)</td>
<td>11.6</td>
<td>14.3</td>
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<td>C1 - Redox Eh (mV)</td>
<td>71.7</td>
<td>46.5</td>
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<tr>
<td>C2 - Redox Eh (mV)</td>
<td>-216</td>
<td>12</td>
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</tbody>
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References


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