Paleo-environmental imprint on microbiology and biogeochemistry of coastal quaternary sediments


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Abstract

To date, North Sea tidal flat sediments have been intensively studied down to a depth of 5 m below sea floor (mbsf). However, little is known about the biogeochemistry, microbial abundance, and activity of sulfate reducers as well of methanogens in deeper layers. For this study, we hypothesized that the imprint of the paleo-environment is reflected in current microbiogeochemical processes. Therefore, 20 m-long cores were retrieved from the tidal-flat area of Spiekeroog Island, NW Germany. Two drill sites were selected with a close distance of only 900 meters, but where sedimentation occurred under different environmental conditions: first, a paleo-channel filled with Holocene sediments and second, a mainly Pleistocene sedimentary succession. In general, the numbers of bacterial 16S rRNA genes are one to two orders of magnitude higher than those of Archaea. The abundances of key genes for sulfate reduction and methanogenesis (dsrA and mcrA) correspond to the sulfate and methane profiles. A co-variance of these key genes at sulfate-methane interfaces and enhanced potential AOM rates suggest that anaerobic oxidation of methane may occur in these layers. Microbial and biogeochemical profiles are vertically stretched relative to 5 m-deep cores from shallower sediments in the same study area. Compared to the deep marine environment, the profiles are transitional between the shallow subsurface and the marine deep biosphere. Our interdisciplinary analysis shows that the microbial abundances and metabolic rates are elevated in the Holocene compared to Pleistocene sediments. However, this is mainly due to present environmental conditions such as pore water flow and organic matter availability. The paleo-environmental imprint is still visible but superimposed by these processes.
1 Introduction

Tidal flats are coastal areas that are characterized by daily flood and ebb currents resulting in highly dynamic changes in salinity, incident light, oxygen availability and temperature. The network of sand flats, mud flats, salt marshes, islands, and tidal channels forms a transition zone between land and sea. Tidal flats are found all over the world, for example in Asia along the Korean coast, in North Africa and North America along the Atlantic coast, and in Europe in the British channel and along the North Sea coast. On a global scale, this type of landscape is of great importance due to its high primary production and organic matter remineralization rates (van Beusekom et al., 1999; Cadée and Hegeman, 2002; D’Andrea et al., 2002; de Beer et al., 2005; Loebl et al., 2007). The water column and underlying sediments are tightly coupled due to bidirectional exchange of nutrients and organic material (Huettel et al., 1998; Rusch et al., 2001; Billerbeck et al., 2006a; Billerbeck et al., 2006b; Beck et al., 2008b).

One of the largest tidal areas of the world extends along the southern North Sea coast. It covers about 9300 km² and stretches more than 500 km along the coastline of The Netherlands, Germany, and Denmark (Dittmann, 1999). Within the frame of two long-term research projects (Dittmann, 1999; Rullkötter, 2009), biogeochemical and microbial dynamics in pore waters and sediments were studied in detail in the tidal flat system of Spiekeroog Island (Böttcher et al., 1997; Böttcher et al., 1998; Volkman et al., 2000; Llobet-Brossa et al., 2002; Rütters et al., 2002; Köpke et al., 2005; Billerbeck et al., 2006a; Billerbeck et al., 2006b; Wilms et al., 2006a; Wilms et al., 2006b; Beck et al., 2008a, b, c; Fichtel et al., 2008; Freese et al., 2008; Gittel et al., 2008; Røy et al., 2008; Al-Raei et al., 2009; Jansen et al., 2009). The studies focused on surface sediments as well as on deeper sediment layers down to 5 ms below sea floor (mbsf). In addition, biogeochemical and microbial processes were studied in the water column and on suspended particulate matter (Liebezeit et al., 1996; Grossart et al., 2004; Lunau et al., 2006; Dellwig et al., 2007; Grunwald et al., 2007, 2009; Rink et al., 2007; Stevens et al., 2007; Kowalski et al., 2009). These comprehensive studies render the
tidal flat area of Spiekeroog Island into one of the presently best understood tidal flat ecosystem worldwide.

In the Spiekeroog tidal flat area, sulfate reduction and methanogenesis are the main terminal pathways of anaerobic organic matter remineralization in sediments of 5 m depth. Sulfate-reducing bacteria form highly abundant and active populations in deep sandy sediments (Ishii et al., 2004; Gittel et al., 2008). Sulfate and methane show inverse depth profiles, with a methane maximum in the sulfate-depleted zone (Wilms et al., 2007; Beck et al., 2009). The population sizes of Bacteria, Archaea, sulfate reducers, and methanogens correspond well to the vertical sulfate-methane profiles (Wilms et al., 2007). The activity of sulfate reducers and methanogens and thus the turnover of organic matter are highly influenced by sedimentation rate and pore water flow (Beck et al., 2009). Profiles of the key genes for dissimilatory sulfate reduction and methanogenesis suggest potential anaerobic methane oxidation coupled to sulfate reduction at sulfate-methane transition zones (Wilms et al., 2007).

The activity of sulfate reducers and methanogens varies in the tidal flat area, with enhanced turnover rates at tidal flat margins (Beck et al., 2009). Large areas in the central parts of tidal flats show a low microbial activity. The highest turnover is found close to the low water line and gradually decreases across tidal flat margins within the first tens of meters. This is due to the high availability of fresh organic matter and electron acceptors in margin sediments induced by high sedimentation rates and/or pore water flow. Overall, tidal flats and creeks were identified as a major source for methane emission (Middelburg et al., 2002).

Despite these earlier findings, little is known about the pore water biogeochemistry, microbial abundance, and activity of sulfate reducers and methanogens in sediment layers below 5 mbsf. Wilms et al. (2006a) presented the first evidence for typical deep-biosphere representatives such as the Chloroflexi in relatively young subsurface tidal flat sediments. They concluded that tidal flats might offer the opportunity to study and understand certain processes of the deep biosphere at more easily accessible sites. Common features of subsurface tidal flat sediments and the deep marine biosphere
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were recently summarized by Engelen and Cypionka (2009). Although on different temporal and spatial scales, both environments exhibit similar biogeochemical profiles and microbial successions.

For this study, we hypothesized that the geological history of sediment accumulation and thus the paleo-environment has an impact on microbial abundance, microbial activity, and pore water biogeochemistry in Quaternary deposits. We wanted to test whether the paleo-environmental imprint is still reflected in the microbiology and biogeochemistry. As the sediments are not isolated in space and time from the surrounding environment, biogeochemical processes have continued since sediment deposition. To test our hypothesis, two geological settings were selected as drilling sites which are located close to each other but differ in sediment age and paleo-environmental conditions: A paleo-channel filled with mainly Holocene sediments and a sedimentary succession with the oldest sediments deposited during the Saalian glaciation ca. 130,000 years ago.

2 Material and methods

2.1 Sample collection

This study was carried out in the tidal flat area of Spiekeroog Island, NW Germany. In this region, the base of the Holocene sedimentary succession is characterized by former river valleys which drained towards the north and reached down to 20 m below the modern sea level (Streif, 1990). Based on stratigraphic information given by Sindowski (1972), two drill sites were chosen in the western tidal flat area of Spiekeroog Island (Fig. 1). In November 2007, two 20 m long sediment cores were drilled by percussion core drilling using 1 m long polycarbonate liners (inner diameter 10 cm). Core JS-A was drilled in a deep channel incised in Pleistocene sediments and filled with Holocene material (N 053° 43.563′, E 007° 41.249′). Core JS-B, taken at a distance of 900 m from JS-A on the open tidal flat, comprises sediments of the last two glacial/interglacial cy-
cles (N 053° 43.385′, E 007° 42.011′).

After recovery, the core sections were immediately sealed on the drilling rig. If core sections were not completely filled, a spacer was inserted into the liners to stabilize the core material. As potential contamination with seawater could not be ruled out completely for sandy sections, samples with suspicious chloride and sulfate values were disregarded.

Pore water for measuring chloride, sulfate, sulfide, nutrients, alkalinity, and dissolved organic carbon (DOC) was retrieved from closed core sections using Rhizon samplers (Rhizosphere research products, Wageningen, The Netherlands; Seeberg-Elverfeldt et al., 2005) which were inserted through small drill holes. Additionally, holes with a diameter of 1 cm were drilled into the liners to collect 2 cm³ sediment samples for methane measurements. After pore water and gas sampling, cores were split lengthwise to describe the lithology and to sample sediment for further petrographical, biogeochemical and microbiological analyses.

2.2 Biogeochemical and sedimentological analyses

Pore water sulfate and chloride were measured by ion chromatography with conductivity detection (Dionex DX 300). Ammonium and silica were determined photometrically (Grasshoff et al., 1999). Total alkalinity was determined by a spectroscopic method proposed by Sarazin et al. (1999). The analytical procedures as well as precision and accuracy of the analyses are described in detail in Beck et al. (2008b).

Dissolved organic carbon (DOC) was analyzed with a multi N/C 3000 instrument (Analytik Jena AG, Jena, Germany). Pore water was 5- to 10-fold diluted with purified water, acidified with 2 N HCl and purged with synthetic air to remove inorganic carbon. For calibration, a dilution series of a DOC standard (CertiPUR®, Merck, Germany) was run parallel to the pore water samples.

For sulfide analyses, pore water aliquots were preserved in 5% Zn acetate solution after pore water extraction to prevent sulfide oxidation. Sulfide measurements were carried out by the methylene blue method (Cline, 1969) using a Specord 40 spec-
trophotometer (Analytik Jena AG, Jena, Germany).

For measuring methane concentrations, 2 cm$^3$ of sediment were transferred to gas-tight tubes filled with 20 ml sodium hydroxide solution (2.5%). Headspace samples were analyzed on a Varian CX 3400 gas chromatograph (Varian, Darmstadt, Germany) equipped with a plot-fused silica column (No. 7517; 25 m by 0.53 mm, Al$_2$O$_3$/KCl coated; Chromopack, Middleburg, The Netherlands) and a flame ionization detector.

To analyze the fractions of acid-volatile sulfur ($\text{AVS} = \text{FeS} + \text{H}_2\text{S} + \text{HS}^-)$ and chromium reducible sulfur ($\text{CRS} = \text{FeS}_2 + \text{S}^0$), wet sediments were sub-sampled with cut plastic syringes and immediately fixed with Zn acetate solution (20%). A two-step distillation procedure (Fossing and Jørgensen, 1989) was applied to extract the AVS and CRS fractions separately.

The composition and texture of iron sulfides was identified by electron microscopy (Quanta 400 Scanning Electron Microscope, FEI, Oregon, USA). Dried samples and an acceleration voltage of 15 kV were used for analyses.

Sedimentary structures of the cores were described by visual inspection. Grain size analysis was performed on fresh, ultrasonically suspended samples by means of a Fritsch Analysette 22 laser granulometer (Idar-Oberstein, Germany). Coarse material $>500$ µm was removed before analysis. The porosity was calculated using an empirical relationship between water content and porosity determined for a large number of samples from the same area. Water contents were determined by weighing before and after freeze drying. Total carbon and sulfur were analyzed by combustion in a CS 500 IR element analyzer (Eltra, Neuss, Germany). The total organic carbon content (TOC) was determined as the difference between total carbon and inorganic carbon measured with a CO$_2$ coulometer (UIC, Joliet, IL). For precision and accuracy of carbon determinations see Babu et al. (1999).

### 2.3 Total cell counts

Total cell counts were obtained by means of SybrGreen I as fluorescent dye according to a protocol of Lunau et al. (2005), which was adapted to sediment samples. For sam-
ple fixation, 1 cm$^3$ of sediment was transferred to 10 ml of a glutaraldehyde solution (1%) immediately after opening the cores. One milliliter of the homogenized sediment slurry was buffered with 2.5 µl of 50 × TAE buffer. After one or two days of incubation, the samples were washed twice with TAE buffer. For detaching cells from particles, 300 µl methanol were added. The suspension was incubated for 15 min at 35°C in an ultrasonic bath (Bandelin, Sonorex RK 103 H, 35 kHz, 2x320 W per period). Homogenized aliquots of 20 µl were equally dispensed on a clean microscope slide in a square of 20 × 20 mm. The slide was dried on a heating plate at 40°C. A drop of 12 µl staining solution (190 µl Moviol, 5 µl SybrGreen I, 5 µl 1 M ascorbic acid in TAE buffer) was placed in the center of a 20 × 20 mm coverslip, which was then placed on the sediment sample. After 10 min of incubation, 20 randomly selected fields or at least 400 cells were counted for each sediment sample by epifluorescence microscopy.

2.4 DNA extraction and quantification

Total genomic DNA was extracted from 0.5 g sediment by using the FastDNA SpinKit (Q-BIOgene, Carlsbad, Canada) according to the manufacturer’s instructions. DNA concentrations were quantified fluorometrically in a microtiter plate-reader (FLUOstar Optima, BMG Labtechnologies, Offenburg, Germany) using PicoGreen (Molecular Probes, Eugene, USA). For quantification, 100 µl of PicoGreen solution (diluted 1:200 in TE buffer; pH 7.5) was added to 100 µl of DNA extract (each sample diluted 1:100 in TE buffer; pH 7.5) and subjected to a microtiter plate. Serial dilutions of Lambda-DNA (100 ng µl$^{-1}$ to 1 ng µl$^{-1}$) were treated as described above and served as a calibration standard in each quantification assay. Fluorescence was measured at an excitation of 485 nm and an emission of 520 nm.

2.5 Quantitative PCR

The abundance of gene targets that are molecular markers for different phylogenetic and physiological groups of microorganisms was determined by quantitative (real-time)
PCR. In detail, a modified protocol according to Wilms et al. (2007) was used to quantify the domains *Bacteria* and *Archaea*, as well as key-genes for sulfate reduction and methanogenesis, the α-units of the dissimilatory sulfite reductase (*dsrA*) and the methyl coenzyme-M reductase (*mcrA*), respectively. The latter is also used as a signature gene for anaerobic methane oxidation. All primer sequences are given in Wilms et al. (2007). Before quantification, serial dilutions of standard DNA had to be prepared for calibration. *Desulfovibrio vulgaris*<sup>T</sup> (DSM 644) was chosen as the standard organism for the bacterial 16S rRNA gene and the *dsrA* gene. A culture of *Methanosarcina barkeri*<sup>T</sup> (DSM 800) was used as standard for the quantification of the archaeal 16S rRNA gene and the *mcrA* gene. Genomic DNA was extracted from liquid cultures using the FastDNA SpinKit (Q-BIOgene, Carlsbad, Canada). These DNA extracts were used to generate calibration standards for the quantification of the *dsrA* gene and the *mcrA* gene. Considering that one genome harbors only one of the key genes, the standards ranged from 4.4 × 10<sup>5</sup> to 4.4 × 10<sup>0</sup> gene copies per µl. To generate standards for quantifying *Bacteria* and *Archaea* the almost complete 16S rRNA genes were amplified prior to quantitative PCR by using the primer pairs 8f/1492r and 8f/1517r, respectively. Amplification conditions for the bacteria-specific PCR were described by Süss et al. (2004), those for the archaea-specific approach by Vetriani et al. (1999). The PCR amplicons were diluted from 1 × 10<sup>8</sup> to 1 × 10<sup>0</sup> gene copies per µl. The calibration standards were additionally amplified in every qPCR run for quantification.

Quantitative PCR amplification was performed in 25 µl volume containing 12.5 µl of the DyNAmo<sup>TM</sup> HS SYBR Green qPCR Kit (Finnzymes Oy, Espoo, Finland), 0.2 mM of each primer, 1.5 µl PCR-water and 10 µl of the 1:100 diluted DNA templates. Thermal cycling was performed by using a Rotor-Gene, RG-3000 four channel multiplexing system (Corbett Research, Sydney, Australia) with the following parameters: 95°C initial hold for 15 minutes to activate the *Taq* polymerase, followed by 50 cycles of amplification, with each cycle consisting of denaturation at 94°C for 10 s, followed by 30 s annealing at primer-specific temperatures and the elongation step at 72°C for 30 s. Fluorescence was measured at the end of each amplification cycle for 20 s at 82°C.
After each run, melting curves were recorded to ensure that only specific amplification had occurred.

### 2.6 Spiking experiments to determine the DNA recovery rate

Five sediment samples per core were selected to determine the DNA recovery rate prior to gene target quantification by qPCR (app. depth: 0.5, 4, 9, 14 and 18 mbsf). The samples comprised different lithological structures of intertidal sands, marine mud, fluvial clays and fluvial sands. The samples (0.5 g) and a glass-bead control were spiked with $1 \cdot 10^9$ cells of *Rhizobium radiobacter*. The DNA recovery rate was determined after DNA extraction and quantification of the bacterial 16S rRNA gene as described above.

### 2.7 Ex situ rates of anaerobic oxidation of methane (AOM) and sulfate reduction (SRR)

For rate measurements of AOM and SRR, triplicate samples of each sediment horizon were collected in glass columns ($\sim$8 ml), which were immediately closed with butyl rubber stoppers (Niemann et al., 2005). The samples were stored in argon-flushed Ziploc bags and kept at 8°C. In the laboratory, $^{14}$C methane ($\sim$1 kBq) and $^{35}$S sulfate (250 kBq) tracers were injected through the stopper. After incubation for 96 h, AOM incubations were stopped by transferring the samples to 50 ml glass vials filled with 25 ml sodium hydroxide (20% w/w). All glass vials were shaken thoroughly to equilibrate the pore water methane in the headspace and the sediment. After 76 h of incubation, SRR samples were transferred to 50 ml Falcon tubes filled with 20 ml (20% w/w) zinc acetate solution. Additionally, dead controls were prepared for AOM and SRR, where the tracer was added to the samples directly after transferring them to their fixative. Samples were processed and rates were calculated as described by Treude et al. (2003) and Kallmeyer et al. (2004). At some depths, sulfate concentration is interpolated to calculate SRR. Background thresholds for AOM and SRR were controls plus three times standard deviation and rates were only taken into account when at least two of
the triplicate measurements per horizon were above detection limit.

2.8 Modeling

To obtain net sulfate reduction rates, the interstitial water gradients of dissolved sulfate were modeled for steady-state conditions using the one-dimensional modeling code PROFILE (Berg et al., 1998) as described previously (Böttcher et al., 2004). Diffusion coefficients were corrected for in-situ temperature and salinity according to Schulz (2006) and Boudreau (1997).

3 Results

3.1 Sediment structure and paleo-environment

Core JS-A (Fig. 2a) was drilled into a paleo-channel filled with Holocene sediment (Sindowski, 1972). Its uppermost part consists of fine-grained, well-sorted intertidal sand with low contents of mud (i.e. <20% in the grain size fraction <63 µm) and occasional mollusk shells. The porosity is approximately 35%. The organic carbon content is generally low (<0.2%), but reaches up to 0.9% TOC in a few thin brownish layers.

The channel fill sediments between 3.5 and 12.2 mbsf consist of grey mud containing >85% silt with little fine sand. Thin, distinct layers of very fine sand occur in the lower part of this section, but are rare in the upper part. Both top and bottom of this interval are erosive. The carbon content ranges between 1 to 2.4%. Due to the fine grain size, the porosity is high (avg. 56%) and permeability is low.

The underlying Pleistocene fluvial fine sands (12.2-18.4 mbsf) contain minor amounts of mud or coarse sand. They are poor in organic carbon (<0.2%). However, peat fragments, few thin layers of organic material and dispersed plant particles were found. The ages of these sands are not clearly constrained. According to Sindowski et al. (1972), a Saalian age (ca. 300 000–128 000 years) is likely for the lower
part of this section, but an Eemian age (ca. 128 000–115 000 years) is possible for the upper part of the fluvial sediments. Drilling of JS-A was stopped at 18.4 mbsf in bluish green, strongly compacted clay and fine silt (1.6% TOC, ca. 22% porosity). This material belongs to the Lauenburg Clay, a ca. 325 000 year old basin sediment of the Elsterian glaciation.

Core JS-B (Fig. 2b) extends into the Saalian glaciation more than ca. 130 000 years ago. The core has a 30 cm thick surface layer of fine, shell- and mud-bearing intertidal sand underlain by Holocene brownish to medium grey and black mottled mud (Fig. 2). The presence of roots and thin shell and sand layers suggests deposition in an inter- to supratidal mud flat or salt marsh environment. Grain size distributions show two broad maxima in the fine silt (5–8 µm) and the very coarse silt range (30–60 µm). According to Chang et al. (2007), these distributions represent a mixture of fine, aggregated silty clay with a coarser sorted silt fraction. The mud contains about 1% TOC and has a porosity of ca. 55%. The base of the mud sediments at 4.6 mbsf is marked by a 12 cm thick layer of fen peat with a $^{14}$C age of 5490 (±40) years BP.

Between 4.7 and 9.3 mbsf, core JS-B is dominated by light brown to grey silty fine sand and poorly sorted sandy silt of Wechselian age (ca. 115 000–10 700 years). The upper part contains organic matter and some plant remains. Between 9.3 and 14.1 mbsf, sediments of the Eemian interglacial (ca. 128 000–115 000 yrs) are present. From bottom to top, the Eemian sediments show a transgressive sequence with a solid layer of peat with large wood fragments and leaf remnants at the base. This is overlain by grey-brown, homogeneous, almost sand-free silt and clay with 1.2 to 3.3% TOC deposited under marine and low-energy conditions. The top of this silt and clay interval is eroded. The overlying regressive sequence consists of beige-colored, fine-grained, probably coastal sands with variable admixtures of silt (6–30%). Layers of medium to coarse sand occur near the bottom and top. The sands are followed by lacustrine, light to medium brown mud with thin layers or lenses of organic matter (TOC 1.6–3%) from 9.3 to 11.1 mbsf. The lacustrine and marine mud units are clearly distinguished by different organic carbon/sulfur ratios, $^{13}$C content of the organic matter, the amount and
texture of iron sulfides (especially pyrite) and micro-paleontological evidence (Freund, Köster, Böttcher, unpublished data). The deepest part of the core from 14.1 mbsf to the bottom is formed by fluvial sand of the Saalian glacial period (ca. 478 000–128 000 yrs). It consists of beige- to grey-colored fine- to medium-grained sand. The fine fraction is mainly coarse silt and makes up 5-18%. Admixtures of coarse sand and charcoal particles are common.

### 3.2 Biogeochemistry

At both sites, pore water salinities drop with depth from seawater values of 30 in the upper layers to brackish values around 10 indicating saltwater-freshwater mixing (Fig. 3). Salinities are lowest near the boundary between Pleistocene and Holocene deposits due to the freshwater-dominated water regime at this time of sedimentation. Below, salinities increase again which implies the presence of a deep saltwater reservoir.

Similar to chloride, sulfate concentrations sharply decrease within the upper layers (Fig. 3). Nevertheless, the decrease in sulfate is not only due to saltwater-freshwater mixing, but also due to microbial sulfate reduction. Sulfate consumption is most pronounced at 5 to 11 mbsf in core JS-A, where sulfate shows a depletion of 14 mM after correcting for changes in chloride. Sulfide concentrations display pore water maxima in the zones of net sulfate reduction, e.g. within the top 6 m of core JS-A and the upper 12 m of core JS-B. However, part of the sulfide produced by sulfate reduction may have been lost to the sediment, most likely due to iron sulfide formation or interaction with organic matter in peat layers.

Reduced sulfur accumulated in the sediments as iron sulfides (Fig. 3). The pyrite contents (CRS) almost exactly mimic the lithological changes (Fig. 2) indicating that the very early diagenetic sulfur cycle was responsible for sulfide retention. Enhanced TOC contents in mud-rich sediments likely provided the substrate for sulfate-reducing bacteria. The occurrence of pyrite mostly in form of framboids (Fig. 4) supports the interpretation of pyrite formation close to the sediment-water interface (Wilkin et al., 1996). Euhedral pyrite in both cores shows that pyrite formation still went on af-
ter deeper burial. In contrast, the AVS (FeS and dissolved sulfide) content is not as strongly related to lithology, but follows the zones of enhanced microbial activity.

Methane concentrations are fairly low in sediment layers where sulfate is still present and are rarely exceeding 10 nmoles per gram of sediment (Fig. 3). Due to the competition for electron donors and the fact that sulfate reduction is energetically favored over methanogenesis, methanogens are probably outcompeted by sulfate reducers in these layers. Only where sulfate is almost depleted, methane is found with a maximum of 350 nmol per gram of sediment (between 5 and 12 mbsf in core JS-A).

The products and indicators of organic matter remineralization such as alkalinity, DOC and ammonium show parallel trends (Fig. 3). Highest concentrations of these compounds are observed at intermediate salinities of 15 to 20 within the Holocene silt interval of core JS-A, which also hosts elevated amounts of organic matter. The DOC concentrations and the alkalinity values are therefore most likely related to the availability of degradable substrate and its degradation by sulfate reduction. The reverse covariation of alkalinity and sulfate indicates a connected process.

Dissolved silica concentrations cover a broad range from 40 to 1400 µm (Fig. 3). Highest concentrations are associated with the Holocene salt marsh sediments. Silica enrichment in marine and estuarine pore waters results from the dissolution of biogenic silica after deposition and burial of diatom frustules (Dixit and van Cappellen, 2002; Gallinari et al., 2002). Dissolved silica may therefore serve as a chemo(litho)stratigraphic marker to distinguish marine from non-marine deposits. Diffusional transport across lithological boundaries, however, must be taken into account.

### 3.3 Microbial abundance

The depth profiles of bacterial and archaeal 16S rRNA genes obtained by quantitative (real-time) PCR generally follow the total cell counts (Fig. 5). While core JS-B exhibits constant depth profiles for all quantifications, a drop by one order of magnitude was found in JS-A at 11 mbsf. This coincides with the lithological boundary between Holocene mud-rich and Pleistocene sandy sediments. In general, the bacterial 16S
rRNA gene copy numbers are one to two orders of magnitude higher than those of *Archaea*. The distribution of *mcrA* genes corresponds to the numbers of archaeal 16S rRNA genes in both cores. In contrast to core JS-B, core JS-A exhibits a high proportion of *mcrA* genes in comparison to the number of archaeal 16S rRNA genes. This suggests that most of the *Archaea* in core JS-A harbor the molecular signature gene for methanogenesis or anaerobic methane oxidation while most of the *Archaea* in core JS-B are probably not capable of these processes.

A similar trend was observed for the *dsrA* gene, the molecular marker for sulfate reduction. While the numbers of dsrA genes are one to three orders of magnitude lower than the numbers of bacterial 16S rRNA genes in core JS-A, they are up to four orders of magnitude lower in core JS-B. This suggests that at site JS-A a higher proportion of bacteria is able to use sulfate as electron acceptor for anaerobic respiration than in core JS-B.

The abundances of the described gene targets corresponds well with the sulfate and methane profiles along the depth profiles of the cores. At site JS-A, increasing numbers of bacterial 16S rRNA genes and *dsrA* genes were detected around the two sulfate-methane transition zones (Fig. 6). The distribution of *mcrA* genes and archaeal 16S rRNA genes shows similar peaks with depth. The co-occurrence of both key genes for anaerobic oxidation of methane at the expense of sulfate suggests that anaerobic oxidation of methane may take place at the sulfate-methane transition zones. In core JS-B, the relationship of sulfate to the number of *dsrA* genes is reflected by their numerical increase around 4 mbsf. Further, the sulfate minimum matches the elevated sulfide values (Fig. 3) suggesting the presence of an active sulfate reducing community in this layer.

### 3.4 Microbial activity

At site JS-A, potential sulfate reduction rates are highest in the upper 5 mbsf where sulfate is available at almost seawater concentrations (Fig. 6). Below, sulfate reduction rates are diminished due to the lack of sulfate. However, below 12 mbsf, where sulfate
reappears, it is reduced at rates similar to those in the upper 5 mbsf. At site JS-B, potential sulfate reduction rates are highest in the Holocene sediment layers down to about 4 mbsf. Significant, but low rates were also measured at 9, 13 and 18 mbsf, whereas in the intervals between these horizons sulfate reduction rates were below the detection limit.

Modeled net sulfate reduction rates from pore water sulfate profiles are ten to hundred times lower compared to gross rates reported for highly active surface sediments in the study area (Al-Raei et al., 2009; Beck et al., 2009). Modeling the top 6 m of core JS-A and the top 2.7 m of core JS-B yields depth-integrated sulfate reduction rates of 0.1 and 0.3 mmol m\(^{-2}\) d\(^{-1}\), respectively (considering a pelagic sulfate concentration of 26 mM). In addition, pore water modeling gave volumetric rates of 1.2 nmol cm\(^{-3}\) d\(^{-1}\) (JS-B) to 16 nmol cm\(^{-3}\) d\(^{-1}\) (JS-A), which are higher than the gross rates measured with radio tracers (Fig. 6). This suggests that sulfate reducers may be stimulated to a different extent during rate measurements under laboratory conditions.

Potential anaerobic methane oxidation rates peak at about 5 mbsf and 11 mbsf at site JS-A (Fig. 6). This fits nicely with the sulfate-methane transition zones and the slightly increased numbers of mcrA and dsrA gene copies close to these interfaces. In contrast, AOM rates are very low at site JS-B. This may be due to the low methane concentration in this core where sulfate is still present at all depths.

4 Discussion

4.1 Imprint of paleo-environmental conditions

Microbial abundance and activity (Figs. 5, 6) are linked to paleo-environmental conditions via sediment structure, organic matter quality and geochemical composition. At present, potential sulfate reduction rates in both cores are highest in the uppermost layers where sulfate and most likely labile organic material are available for anaerobic respiration. We speculate that the sediments, which are presently found at 5 to
11 mbsf in core JS-A, also exhibited higher sulfate reduction rates during the time of deposition. This assumption is supported by the almost complete depletion of sulfate, the still elevated TOC content as well as the framboidal structure of pyrites. Even if euhedral pyrite structures show that pyrite formation still occurred after deeper burial of the sediments, the sulfur cycle appears to be primarily controlling microbial and thus biogeochemical dynamics in freshly deposited sediments, i.e. under the environmental conditions prevailing during deposition.

Conclusions about the paleo-microbial activity are, however, hampered by better preservation of microbial and biogeochemical signals in certain lithological layers. For example, at site JS-A this preservation is indicated by the enrichment of remineralization products within mud-dominated sediments in the depth interval of 5 to 11 mbsf. The current sulfate reduction rates measured in these layers are much lower due to the lack of sulfate and labile organic matter (Fig. 6).

Although sediment age has an impact on microbiology in core JS-B, the imprint of the paleo-environment on microbial processes is less pronounced than expected. We did not find a relationship of lithology with microbial abundance and activity as described e.g. for ash layers in comparison to clay layers in coastal subseafloor sediments (Inagaki et al., 2003). Rather the lithological structure influences the exchange between different layers due to diffusion and possibly advection barriers. Furthermore, increasing sediment age may have led to decreasing bioavailability and increasing recalcitrance of organic matter. Thus, the Eemian mud-rich intervals with elevated TOC contents do not leave an imprint on the current microbial activity. This results in a homogeneous distribution of more or less inactive microbial communities along the depth profile of core JS-B.

4.2 Imprint of current environmental conditions

The imprint of the paleo-environment on microbiology and biogeochemistry may be reduced by environmental conditions that have developed more recently. Our results provide hints that the current hydrogeological situation influences biogeochemistry and
microbial activity. Seawater intrusion in deep permeable sands may transport sulfate to formerly sulfate-poor layers (Fig. 2, 3). Deep saltwater encroachments are common in coastal aquifers and are typically explained as re-circulating seawater (Cooper, 1964; Kohaut, 1964; Reilly and Goodman, 1985; Reilly, 1993). At site JS-A, the re-supply of sulfate leads to enhanced sulfate reduction rates in Pleistocene sands below 12 mbsf, reaching almost the same order of magnitude as those in the uppermost meters (Fig. 6). In contrast, no sulfate is replenished in the mud-rich layers from 5 to 11 mbsf where sulfate concentrations are close to zero and thus very low current sulfate reduction rates prevail. Former studies down to 5 m depth indicated that microbial processes in central parts of the Janssand tidal flat are only slightly influenced by pore water flow (Beck et al., 2009). Nevertheless, flow velocities may be fast enough to re-supply sulfate to sites JS-A and JS-B in the central part over long time scales such as several thousands of years.

At site JS-B, the hydrogeological setting and the microbial response are different. The lithological barrier at about 4 mbsf marks the lower boundary of the sulfate reduction zone. The weakly permeable Holocene mud layers inhibit the downward supply of sulfate and dissolved organic material to the Pleistocene sand-dominated layers below. The deep saltwater intrusion described above is also visible in the chloride and sulfate profiles of core JS-B. However, the shapes of the profiles indicate a diffusion-dominated system without microbial sulfate reduction. The current abundance of terminal degraders of organic matter is much lower at site JS-B compared to site JS-A, even though we found similar cell counts in both cores (Fig. 5).

Organic matter degradation is most often syntrophically performed with the help of fermenting microorganisms. Due to the high diversity of pathways and organisms involved in these processes, their analysis is highly challenging. Furthermore, fermentation products do not accumulate in the sediment, which makes their monitoring rather difficult. However, terminal degraders of a metabolically active microbial community are fueled by fermenters and enhanced terminal degradation processes would be detectable by the methods applied here. Our findings indicate that most of the detected
microorganisms in core JS-B are presently not active or may even be spores. The latter explanation is supported by viable counts which show that pasteurized incubations exhibit almost the same numbers as non-pasteurized incubations (data not shown). Further, endospore numbers increase from about 1% of total cell counts in Janssand surface sediments to 10% at 5 mbsf indicating an increasing number of bacteria in this resting stage with depth (Fichtel et al., 2008). At site JS-A, pasteurized and non-pasteurized incubations result in similar numbers at 0.5 and 18 mbsf as for site JS-B. However, more viable cells than spores are detected at intermediate sediment depth indicating a more active community.

Spores will probably not contribute to the methodological divergence of direct cell counting and molecular quantification (Fig. 5). They are generally not detected by both approaches due to their cortex preventing staining as well as a sufficient cell disruption for DNA extraction. The difference between total cell counts and molecularly determined numbers of gene copies may have several reasons. First, direct counting by using fluorescent dyes overestimates the actual numbers in most cases due to unspecific staining of sediment particles (Morono et al., 2009). Second, the primers used do not target all sequences and may discriminate certain groups of organisms. Third, DNA extraction from deep seafloor sediments is always characterized by a low efficiency (D’Hondt et al., 2004). A spiking experiment with five representative samples per core and a defined number of cells indicated a recovery rate of 33-82% after DNA extraction and qPCR amplification. This rate may even be lower for indigenous microorganisms as many of them are probably attached to particles or protected by exopolymeric substances.

4.3 The transition between shallow tidal-flat sediments and the deep biosphere

The astonishing similarities between certain biogeochemical and microbiological features of the upper 5 meters of tidal-flat sediments and the upper hundred meters of deep-sea sediments were already highlighted by Engelen and Cypionka (2009). They propose the subsurface of tidal flats to be a model for the marine deep biosphere. This
was further supported by the investigations of Beck et al. (2009). Their analysis of an 80 meter transect comprising five cores from the low water line to the top of the Janssand tidal flat suggested stretching of biogeochemical profiles into deeper layers at sites of decreasing microbial activity. The investigated 20 m-long sediment cores fit nicely into this concept, on an intermediate temporal and spatial scale. Sulfate and methane profiles at site JS-A are strikingly similar to several of those described for the deep biosphere (D’Hondt et al., 2004; Parkes et al., 2005; Engelen et al., 2008). The sulfate input from the water column to the surface sediments and the replenishment of sulfate in deep sediment layers results in enhanced activity of sulfate reducers. In previous investigations of app. 5 m-long sediment cores, increased numbers of \( dsrA \) and \( mcrA \) gene copies at the sulfate-methane transition zone led to the suggestion that sulfate-dependent anaerobic oxidation of methane may occur (Thomsen et al., 2001; Wilms et al., 2007). We now can verify this assumption by direct measurements of elevated potential AOM rates determined at the respective layers of the 20 m-long cores studied here.

The intermediate range of microbial activities at sites JS-A and JS-B compared to shallower tidal flat sediments and the deep biosphere is also reflected by the range of sulfate reduction rates. In tidal flat sediments down to 5 m depth, rates are up to three orders of magnitude higher (Beck et al., 2009) than in the 20 m-long cores. These high rates were obtained at sites influenced by pore water flow and/or fast sedimentation. Without such stimulating processes, rates may be substantially lower as observed previously in central parts of tidal flats. Thus, the potential sulfate reduction rates in the 20 m-long core are quite similar to those of the marine deep biosphere (Parkes et al., 2005). The abundance of \textit{Bacteria} and \textit{Archaea} is in the lower range of values typical for the upper 5 meters of the tidal-flat subsurface (Wilms et al., 2007) and in the upper end compared to deep-sea sediments (Engelen et al., 2008).
5 Conclusions

Our interdisciplinary approach has shown that the paleo-environment has less impact on current microbial abundance and activity than hypothesized. Sediment horizons exhibiting different deposition histories do not lead to a specific microbial response. Instead, the lithological succession influences the present hydrogeological regime in terms of availability of electron donors and acceptors. Thus, the paleo-environmental imprint is still visible but superimposed by modern processes.

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Fig. 1. Study sites JS-A and JS-B on the tidal flat Janssand located close to Spiekeroog Island in a tidal area stretching along the coasts of the Netherlands (NL), Germany (D), and Denmark (DK).
Fig. 2. Lithological profiles of the two sediment cores JS-A and JS-B. Core JS-A shows Holocene sediments filling a paleo-channel. Core JS-B displays a mainly Pleistocene sedimentary succession deposited since the Saalian glaciation more than 130,000 years ago. M, S, Sh indicate mud, sand, and shells, respectively. Orange: shell bed, yellow: sand, green: mud, red: peat, diagonally hatched: intervals stained by humous matter, white vertically hatched: core loss. Sediment cores were collected in November 2007.
Fig. 3. Biogeochemistry at sites JS-A (filled circles) and JS-B (open circles) in the subsurface of the Janssand tidal flat: Salinity, sulfate, sulfide, chromium reducible sulfur (CRS), acid volatile sulfur (AVS), methane, dissolved organic carbon (DOC), alkalinity, ammonium, and silica.
Fig. 4. Pyrite crystals from sites JS-A and JS-B. Framboidal pyrite indicates formation upon microbial sulfate reduction close to the sediment-water interface. Differences in framboid sizes are in agreement with dynamic redox conditions and oxic bottom waters (Wilkin et al., 1996). In contrast, the single euhedral crystals were formed within the sediment later upon activity of sulfate reducers (Wang and Morse, 1996).
Fig. 5. Total cell counts and abundances of bacterial and archaeal 16S rRNA gene copies, dsrA and mcrA gene copies at sites JS-A and JS-B. Total cell counts were directly obtained by means of SybrGreen staining, whereas others were obtained by quantitative PCR. Core JS-B exhibits constant depth profiles for all quantifications. In contrast, a drop in all quantifications by one order of magnitude was found below the lithological boundary at 12 mbsf of core JS-A.
Fig. 6. Sulfate and methane concentrations, copy numbers of \textit{dsr}A and \textit{mcr}A genes, potential sulfate reduction rates (SRR), and potential rates of anaerobic oxidation of methane (AOM) at sites JS-A and JS-B. The grey bars highlight the sulfate-methane interfaces. The same scales are used for both sites. Enhanced potential SRR are coupled to higher \textit{dsr}A copy numbers and consumption of sulfate, whereas a higher abundance of \textit{mcr}A genes corresponds to methane concentrations. Site JS-A shows two sulfate-methane transition zones with detectable methane consumption and elevated potential AOM rates.