Dear Dr. Reisdorph,

I agree with reviewers that major changes need to be done before considering publishing in the journal. I find that your responses to them are not very well supported. I am elaborating further on this as follows.

Sincerely yours

Silvio Pantoja
Associate Editor

1. Abstract. Please rewrite sentence starting Line 2, page 4: “Seasonally averaged data were analyzed on a regional basis to account for distinct biogeochemical differences within the bay due to spatial variation in rates of primary production and the influence of glacial-fed stratification, particularly in the northern regions”

AR: This sentence has been deleted.

Explain what you mean with regional basis, or delete. In any case the sentence does not say much “Respiration and air-sea gas exchange were the dominated drivers of carbon biogeochemistry between the fall and winter of 2012.” This is the abstract of your results: Which aspect of “carbon biogeochemistry” are you referring to?

AR: The lines mentioning ‘regional basis’ have been removed. ‘Biogeochemistry’ has been changed to ‘carbon chemistry’.

Line 10, page 4 in abstract and rest of text:
“The highest carbon production occurred within the lower bay between the summer and fall of 2011 with 11 ~1.3x10^10 g C season-1. Bay-wide, there was carbon production of ~2.6x10^10 g C season-1 between the summer and fall”

It is not clear when the highest production occurred since “between season” could be 1 day, or 3 months. In agreement with one of the reviewer, please change the unit of time to day or other, but not season.

AR: NCP calculations have been converted to g C/day (rather than /season)
2. Abstract. Page 4, line 13. It is “dominant” driver, Isn’t?

AR: This has been corrected. “dominated” was changed to ‘dominant’.

3. Text. I agree with reviewers regarding acronyms. They are not necessary and make reading more difficult, which is certainly a goal of the journal. Please remove them, including CRM, BOD

AR: Acronyms (GLBA, LB, CB, EA, WA, CRM, BOD, etc) have been removed. Exceptions to this include well-accepted acronyms of chemical variables (DIC, DO, TA, POC, NCP, AOU).

4. Response: “AR: We discuss the influence of wind mixing, as well as glacial flour, on our NCP estimates throughout the ms and within the new Discussion section. Influences on stratification are discussed near the beginning of the Background section. Internal waves and constrictions are discussed near the beginning of the new Discussion section. Stratification (primarily salinity-driven) is also discussed in this new section. The influence of winds, turbulent and tidal mixing are also mentioned throughout the Discussion in places where these mechanisms are identified to impact DIC, TA, NCP and nutrient concentrations. We have also added additional text to address other caveats and assumptions to consider in regards to our NCP estimates. These additions are throughout the text and can be viewed via the tracked changes.“

The referee points to the effect of those “important limitations and caveats that need to be considered “on your interpretation”, i.e. your error bars in the conclusion. Please address this issue

AR: We have added a “Caveats” section (Section 4.0) that discusses these aspects and how they impact our DIC and NCP values.

5. Consider this comment by Referee 3 “The introduction and background need to be shorten and it should focus on more relevant aspects that i) influence NPP fluxes within Glacier Bay and ii) that better explain the caveats that underlay the methodological approach used (see the general comments above)

AR: The Introduction and Background sections have been shortened. These sections now focus on NCP in Glacier Bay, as well as mention some caveats. Additionally, a Caveats section has been added following these sections.

6. Referee 3 “ The justification of the work is (STILL) poorly presented …”
Editor: Is there a scientific question?

AR: We have more fully explained the reason for this work at the end of the Intro and
Conclusion sections.

7. Referee 1. “Figure 4 needs to be redrawn. No scientific information can be extracted this way numbers on the map without error bars.”
Editor: Consider this comment in the new version

AR: Figures 3 and 4 have been combined into a table that includes error estimates. Areas calculated for each region have also been added to this table.

8. Discussion should be shortened and may be separate into sections for clarity. There is a mix of literature data and your results that it is difficult to follow. Discuss your data referring to your figures and values.

AR: The Discussion has been split into 3 distinct sections and has been shortened for more concise discussion. Most citations have been removed so to only include this study’s data. Those remaining are necessary citations regarding our explanations of the results (i.e. Redfield ratios, carbon overconsumption).

9. Response to Referee 1. “We understand that these global comparisons are important. However, we do not feel it is within the primary focus of this ms. We touch on similar fjords from around the world, but the focus of this ms is to provide a first time estimate of NCP in GLBA. Including GLBA in a more detailed global comparison is out of the scope of this ms and may be more appropriate for a future publication.”
For the journal the issues brought up by Referee 1 are important; a local process of broader significance

AR: We have a section within our Intro that discuss NCP estimates of other glaciated fjords (Norway, Chile). This section is concise so as not to stray too far from the topic of this study, but indicates that data worldwide is scarce and how our estimates fit within these global estimates.

10. “The description of the study area lacks numerical information on bathymetry, areas and salinity distribution. The presentation of the results has to be raised to a level of overview and synthesis from the tedious rounds of descriptive text. Graphics and tables might improve the presentation in this respect. The primary subject of the manuscript, net community production, is assessed on the basis of salinity normalized DIC data. The details of the calculations are not sufficiently described but this reviewer recalls the paper by Friis et al., (1999) on the errors which may be introduced by conventional salinity normalization when the low salinity end-members have significant inorganic carbon concentrations (Friis et al., 2003).”
The article needs error estimates, with propagation. Please address this issue properly
AR: Areas of each region (in m²) have been added to the Table 1. A bathymetric map has been added to address this issue as well. Section 5.1 addresses the spatial and seasonal salinity distributions (and a figure added) and has been shortened per reviewer/editor request (see comments below). Regarding Friis et al., 2003, this paper refers to errors within alkalinity estimate as a result of salinity normalization. However, we did not normalize alkalinity, only DIC as stated in the Methods section. We used the carbonate correction, also described in the Methods section, to account for freshwater influences for our NCP estimates. Additionally, our low salinity/low TA samples correspond to low DIC concentrations as well, not high DIC as described in this paper.

11. “RC: Glacial flour is one of the characteristics of glacial waters. Are there any carbonate minerals in the glacial flour that could affect the DIC determinations?”
AR: Added text to the Methods section to address this comment: “While glacial flour may supply some carbonate minerals to the marine system, influencing DIC and CaCO2 concentrations, we were not able to quantify the amount of glacial flour deposited in the Bay or analyze its composition for this study.”
Editor: This could be an important caveat that needs to be properly addressed, ¿how this could affect your interpretation?

AR: A ‘Caveats’ section (4.0) has been added to address the implications of glacial flour (as well as other factors brought up in various RCs) on our interpretation of NCP.

12. “RC: Seasonal water column DIC concentration changes can be a good approximation to determine seasonal NPP (especially in open ocean). This methodological approach has however important limitations mainly because it is difficult to constrain several processes that can add or take out inorganic carbon from the water column (besides the air-sea exchange of CO2 that has been properly addressed in this paper). Boundary conditions in a highly dynamic environment such a fjord are difficult to constrain. The respiration of allochthonous organic carbon from terrestrial (and maybe to a lesser extent oceanic) origin can severely distort in situ NPP estimations hence its implications need to be better addressed (at least the caveats that need to be considered).”
How could it affect your measurements and interpretation?
“Another important flaw of the paper is the poor consideration of physical processes that drive NPP within Glacier Bay. The interplay between seasonal freshwater fluxes, influence of nutrient laden more oceanic waters and wind, tidal and other type of water column mixing/stratification processes (including internal waves, the impact of constrictions etc.) have been poorly treated”
This aspect needs to be considered

AR: Discussion of terrestrial DOC/DOM has been added to a Caveats section (section 4.0). The interaction between freshwater flux, marine source waters, and wind/tidal/turbulent mixing are discussed within the Discussion as reasons for our nutrient, NCP, and air-sea flux values. Additionally, discussion of internal waves has been added within the Caveats section as well.
13. In Fig. 1 what is summer-fall? Is it summer and fall? Explain
AR: Fig. 1 does not show seasons. Figure 3 and 4 (Fig.1 is a location map) that show NCP values across seasonal transitions (between summer and fall, etc) has been made into a table for clearer understanding.

14. Figure 5 is of poor quality
AR: Dots have been made larger and quality of file has been increased.

15. Use “NO3-“ or “nitrate”, not “NO3” throughout the text
AR: This has been corrected throughout ms.

16. In 2.0 Background. Section 2.0 has to be re-written with relevant referenced information, and shorten. The text is too general to be useful at providing background information about your study site.
AR: The Background has been shortened to include only information pertaining to the study area.

17. Replace “Figure 1” with “Fig. 1” throughout the text
AR: All instances of “Figure” were replaced with “Fig.”

18. Replace “Conductivity-temperature-depth (CTD) data were collected on downcasts with a Seabird 19-plus system” with “Conductivity, temperature and pressure were collected on downcasts with a Seabird 19-plus CTD”
AR: Original sentence has been replaced by suggested sentence.

19. Please revise general writing such as in “1 μmoles kg⁻¹” instead of “1 μmol kg⁻¹”, etc.
AR: All instances of “μmoles” & “mmoles” were replaced with “μmol” & “mmol”, respectively.

20. Page 10, line 18: samples were calibrated? Explain or rewrite lines 18 -21
AR: This sentence was reworded to “Certified reference material, prepared and distributed by Scripps Institute of Oceanography, University of California, San Diego (Dr. Andrew Dickson’s Laboratory), were run daily before sample analysis to ensure accuracy of sample values.”
21. Page 11, “Macronutrient samples”, these are seawater samples for…
AR: ‘Nitrate, phosphate and silicate’ has been added in reference to macronutrient samples.

22. Combine in one sentence lines 18–23 in page 11: “Seasonally averaged atmospheric pCO2 values (µatm) were used (388.4, 388.9, 393.4, 393.8 and 391.8 for summer 2011 through summer 2012, respectively). Seasonally atmospheric pCO2 values were averaged from the monthly averaged Mauna Loa archive found at www.esrl.noaa.gov. Seasonally atmospheric pCO2 values were averaged from the monthly averaged Mauna Loa archive found at www.esrl.noaa.gov.”
AR: These sentences have been combined.

23. Page 14, line 2: Salinity was “the” lowest
AR: ‘the’ has been added before ‘salinity’

24. Page 14, line 18: “Again” the isohalines remained. Remove or replace “again”.
AR: “Again” has been deleted.

25. Page 13, section 4.1. Add a figure and reduce at least one page.
AR: This section (now 5.1) has been shortened to one page and a figure has been added.

26. Section 4.2- Show your data in a figure, and reduce text by app. 50%
AR: This section (now 5.2) has been shortened and a figure has been added.

27. Section 4.3. Show figure(s) and reduce text by half at least.
   Same for Section 4.4
AR: Sections 4.3 and 4.4 (now 5.3 and 5.4) have been shortened.

28. Figure 4. Replace by one that shows spatial variability in NCP
AR: Figure 4 (and Fig. 3) have been replaced by a table that lists values calculated for each region including regional areas, NCPs and error values.
29. Page 22, lines 15-16. This is not a place to repeat a figure caption. Change or remove. Same for Page 24, lines 1-2.

*AR: Both instances have been deleted.*

30. Page 24. Lines 4-7 do not add anything. Remove. Same with lines 8-15. What is the relationship with this study? If it has, it should go in a proper place. This is the results section.

*AR: These lines have been removed.*

31. Please summarize Section 4.6

*AR: Section 4.6 has been shortened. Extraneous language and lines have been removed.*

32. Re-organize the Discussion section, and shorten

*AR: The Discussion has been split into 3 distinct sections and has been shortened for more concise discussion. Most citations have been removed so to only include this study’s data. Those remaining are necessary citations regarding our explanations of the results (i.e. Redfield ratios, carbon overconsumption).*

33. Conclusion should be about your work (this work). Rewrite and shorten

*AR: Conclusion has been shortened and limited to only discuss our work in Glacier Bay.*
Abstract
The impact of deglaciation in Glacier Bay has been observed to seasonally impact the biogeochemistry of this marine system. The influence from surrounding glaciers, particularly tidewater glaciers, has the potential to greatly impact the efficiency and structure of the marine food web within Glacier Bay. To assess the magnitude, spatial and temporal variability of net community production in a glaciated fjord, we measured dissolved inorganic carbon inorganic macronutrients, dissolved oxygen and particulate
organic carbon between July 2011 and July 2012 in Glacier Bay, AK. High net community production rates were observed across the bay (~54 to ~81 \text{mmol C m}^{-2} \text{d}^{-1}) between the summer and fall of 2011. However, between the fall and winter, as well as between the winter and spring of 2012, air-sea fluxes of carbon dioxide and organic matter respiration made net community production rates negative across most of the bay as inorganic carbon and macronutrient concentrations returned to pre-bloom levels. The highest carbon production occurred within the lower bay between the summer and fall of 2011 with \(\sim 1.3 \times 10^{10} \text{g C season}^{-1}\). Bay-wide, there was carbon production of \(\sim 2.6 \times 10^{10} \text{g C season}^{-1}\) between the summer and fall. Respiration and air-sea gas exchange were the dominant drivers of carbon chemistry between the fall and winter of 2012. The substantial spatial and temporal variability in our net community production estimates largely reflect glacial influences within the bay, as melt-water is depleted in macronutrients relative to marine waters entering from the Gulf of Alaska in the middle and lower parts of the bay. Further glacial retreat will likely lead to additional modifications in the carbon biogeochemistry of Glacier Bay with unknown consequences for the local marine food web, which includes many species of marine mammals.

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1.0 Introduction

Glacier Bay lies within the Gulf of Alaska (Gulf of Alaska) coastal ocean and is a pristine glacially influenced fjord that is representative of many other estuarine systems that border the Gulf of Alaska (Fig. 1). Glacier Bay is influenced by freshwater input, primarily from many surrounding alpine and tidewater glaciers. The low-nutrient influx of freshwater into Glacier Bay, which is highest (up to ~40% freshwater in surface waters during the summer; Reisdorph and Mathis, 2014) along the northern regions of the bay, affects the nutrient loading and, thus, biological production and carbon dioxide (CO₂) fluxes within the bay. The southern region of the bay is less affected by this runoff due to distance from the glacial influence and is more influenced by marine waters that exchange through a narrow channel with a shallow entrance sill (~25 m).

Over the past ~250 years, Glacier Bay has experienced very rapid deglaciation, which has likely impacted the biological structure of the bay. As the climate continues to warm, additional changes to this ecosystem and marine population have the potential to impact net community production (NCP) within the bay, with cascading effects through the food web. To better understand the seasonal dynamics of the underlying biogeochemistry in Glacier Bay, we used the seasonal drawdown of the inorganic constituents of photosynthesis within the mixed layer to estimate regional mass flux of carbon and rates of NCP along with air-sea flux rates of CO₂. This approach has been used in other high-latitude regions to assess ecosystem functionality (e.g. Mathis et al., 2009; Cross et al., 2012; Mathis and Questel, 2013), including net community production and carbon cycling.

Previous studies have shown there is wide-ranging variability in rates of primary
production within other glaciated fjord systems, though NCP data within these ecosystems are sparse. Fjords within the Central Patagonia region (48°S – 51°S) are strongly influenced by glaciated terrain and freshwater runoff, similar to influences in and around Glacier Bay. A study by Aracena et al. (2011) looked at water column productivity in response to surface sediment export production in various Chilean Patagonia fjords (41°-56°S). They calculated primary production rates during the summer between ~35 mmol C m⁻² d⁻¹ in the more southern regions (52°S - 55°S) and ~488 C m⁻² d⁻¹ to the north (41°S - ~44°S). In Central Patagonia, Aracena et al. (2011) estimated primary productivity at ~57 mmol C m⁻² d⁻¹ in the spring, a value comparable to some seasonal estimates in Glacier Bay, and found primary production rates comparable to those of Norwegian fjords (~9 to ~360 mmol C m⁻² d⁻¹).

There have been a number of studies conducted within Glacier Bay, though conclusions of several studies are contradictory. Many of these studies had a short duration and limited coverage, missing much of the spatial, seasonal, and annual variability (Hooge et al., 2003). This lack of data leads to a significant gap in understanding of carbon cycling in Glacier Bay, as well as a lack of predictability of responses to changes in this estuarine system as climate change progresses. To capture some of the seasonal and spatial variability in the bay, we collected and analyzed monthly sampling data over a two-year period. This sampling regime, along with the variety of samples taken, has provided us with the most robust dataset collected in Glacier Bay and allowed us to elucidate the dynamic nature of NCP in a glaciated fjord. Our goal for this study was to better understand carbon cycling in Glacier Bay and how it is impacted by glacial runoff. Additionally, we wish to fill in some gaps in how these
processes may influence net community production within a glaciated fjord ecosystem, and better understand how continued glacial melt will impact productivity in Glacier Bay, as well as in similar glaciated fjord ecosystems worldwide.

2.0 Background

Glacier Bay was once covered by one large icefield, the Glacier Bay Icefield, that has been rapidly retreating since the Industrial Revolution, scouring the bay and leaving behind many alpine and tidewater glaciers. Currently, the marine portion of Glacier Bay is roughly 100 km from the entrance sill to the end of the west arm, and reaches depths > 400 m and > 300 m in the east arm and west arm, respectively (Fig. 2).

Seasonal variation in factors such as light availability, turbulent or wind mixing and freshwater input, impact physical conditions that are vital to primary production, including stratification, photic depth, and nutrient availability. These drivers of NCP vary temporally and spatially within Glacier Bay. Glacial runoff, along with glacial stream input, impart freshwater into the marine system, especially along the arms of the bay. Peak runoff has been shown to occur during the fall, though there is fairly constant flow from June to September (Hill, 2009). Low-nutrient glacial runoff is prevalent, and while it aids in stratification, its low macronutrient concentrations dilute available nutrients in the northern regions nearest tidewater outflows. In the lower parts of the bay, glacial influence is lower and macronutrients are more abundant allowing higher levels of primary production during spring and summer. Glacier Bay maintains relatively elevated phytoplankton concentrations throughout the year compared to levels observed in similar Alaskan fjords (Hooge & Hooge, 2002). However, insufficient research has been done on...
the biological system within Glacier Bay to understand why this occurs.

For this paper, we have calculated seasonal NCP and air-sea carbon flux for the four regions within Glacier Bay in order to better understand ecosystem production in a glacially dominated environment, representative of much of the southern coastal AK region. This study has greatly enhanced our understanding of how glacial melt and air-sea flux impacts DIC concentrations, and thus NCP, in estuaries, like Glacier Bay, which are numerous along the Gulf of Alaska coast in Alaska, as well as other glaciated fjords worldwide.

3.0 Methods

Ten oceanographic sampling cruises took place aboard the National Park Service’s R/V Fog Lark between July 2011 and July 2012. Water column samples were collected at six depths (2, 10, 30, 50, 100 m and near the bottom) at each station throughout the bay (Fig. 1) with a maximum depth within the west arm of ~430 m (Fig. 2). Sampling depths correspond with those currently being used by the Glacier Bay long-term monitoring program and determined by the USGS in the 1990s. Each ‘core’ station (Fig. 1) was sampled during every oceanographic sampling cruise, while all 22 stations were sampled during the months of July and January. “Surface” water refers to water collected from a depth of 2 m. unless otherwise stated. Seasonal data was calculated by averaging each measured parameter at each depth for all cruises during the respective seasons. The summer season consists of June, July and August, fall includes September and October; winter is comprised of February and March cruises, and the spring season includes the months of April and May. Data has been averaged regionally within each of
the four regions of the bay (lower bay, central bay, east arm, and west arm) (Fig. 1).

Regional boundaries were selected based on historical and ongoing research in Glacier
Bay. Bathymetry data (Fig. 2) was retrieved from the National Geophysical Data Center.

Conductivity, temperature and pressure were collected on downcasts with a
Seabird 19-plus CTD. Dissolved oxygen (DO) was sampled and processed first to avoid
compromising the samples by atmospheric gas exchange. Samples for DO analysis were
drawn into individual 115 ml Biological Oxygen Demand flasks and rinsed with 4-5
volumes of sample, treated with 1 mL MnCl₂ and 1 mL NaI/NaOH, plugged, and the
neck filled with DI water to avoid atmospheric exchange. Dissolved oxygen was sampled
and analyzed using the Winkler titrations and the methods of Langdon (2010). Samples
were analyzed within 48 hours. Apparent oxygen utilization (AOU) was derived from
observed DO concentrations using Ocean Data View calculations in version 4.6.2
(Schlitzer, 2013).

DIC and total alkalinity (TA) samples were drawn into 250 mL borosilicate
bottles. Samples were fixed with a saturated mercuric chloride solution (200 µl), the
bottles sealed, and stored until analysis at the Ocean Acidification Research Center at the
University of Alaska Fairbanks. High-quality DIC data was attained by using a highly
precise (0.02%; 0.4 µmoles kg⁻¹) VINDTA 3C-coulometer system. TA was determined
by potentiometric titration with a precision of ~1 µmoles kg⁻¹. Certified reference
material, prepared and distributed by Scripps Institute of Oceanography, University of
California, San Diego (Dr. Andrew Dickson’s Laboratory) were run daily before sample
analysis to ensure accuracy of sample values. The VINDTA 3C provides real-time
corrections to DIC and TA values according to in-situ temperature and salinity.

While glacial flour may supply some carbonate minerals to the marine system,
influencing DIC and CaCO₃ concentrations, we were not able to quantify the amount of
glacial flour deposited in the bay or analyze its composition for this study.
Macronutrient samples (nitrate, phosphate, silicate) were filtered through 0.8 µm Nuclepore filters using in-line polycarbonate filter holders into 25 ml HDPE bottles and frozen (-20°C) until analysis at UAF. Samples were filtered to remove any particles, such as glacial silt, that had the potential to clog equipment during analysis. Samples were analyzed within several weeks of collection using an Alpkem Rapid Flow Analyzer 300 and following the protocols of Mordy et al. (2010).

Particulate organic carbon (POC) samples were collected from Niskins into brown 1 L Nalgene bottles and stored for filtering within 2 days of collection. Samples were collected at 2 m, 50 m and bottom depths. A known volume of samples was filtered through muffled and preweighed 13 mm type A/E glass fiber filters using a vacuum pump. Muffling involved using tweezers to wrap filters in aluminum foil and heating them at 450°F for ~6 hours in a muffling furnace in order to remove any residual organic material. Filtered samples were frozen for transport back to UAF where they were then dried and reweighed. Analyses were completed by OARC at UAF and were run using the methods outlined in Goñi et al. (2001).

The partial pressure of CO₂ (pCO₂) was calculated using CO2SYS (version 2.0), a program that employs thermodynamic models of Lewis and Wallace (1995) to calculate marine carbonate system parameters. Seasonally averaged atmospheric pCO₂ values (µatm) were used (388.4, 388.9, 393.4, 393.8 and 391.8 for summer 2011 through summer 2012, respectively, and were averaged from the monthly averaged Mauna Loa archive found at www.esrl.noaa.gov. For seawater pCO₂ calculations in CO2SYS we used K₁ and K₂ constants from Mehrback et al., 1973 and refit by Dickson and Millero (1987), KH₂SO₄ values from Dickson, the seawater pH scale, and [B]₁ value from
CO₂ fluxes were calculated using seasonally averaged seawater temperature, wind speed, and seawater and atmospheric pCO₂ data using the equation,

\[ \text{Flux} = L \times (\Delta p_{CO₂}) \times k \]  
(Eq. 1)

where \( L \) is the solubility of CO₂ at a specified seawater temperature in mmol m⁻³ atm⁻¹ and \( \Delta p_{CO₂} \) represents the difference between seawater and atmospheric pCO₂ in µatm. \( k \) is the steady/short-term wind parameterization in cm hr⁻¹ at a specified wind speed and follows the equation,

\[ k = 0.0283 \times U^3 \times \left( \frac{Sc}{660} \right)^{-1/2} \]  
(Eq. 2)

where \( U \) is wind speed in m s⁻¹, \( Sc \) is Schmidt number, or the kinematic velocity of the water divided by the molecular diffusivity of a gas in water, and was normalized to 660 cm hr⁻¹, equivalent to the \( Sc \) for CO₂ in 20°C seawater (Wanninkhof and McGillis, 1999). Wind speeds were cubed using the methods of Wanninkhof and McGillis (1999) in an attempt to account for the retardation of gas transfer at low to moderate wind speeds by surfactants and the bubble-enhanced gas transfer that occurs at higher wind speeds.

Seawater temperatures for flux calculations were taken from surface bottle CTD data. Wind speeds were obtained from a Bartlett Cove, AK weather station (Station BLTA2) located in Glacier Bay and maintained by the National Weather Service Alaska Region.

NCP calculations were made using the seasonal drawdown of photosynthetic reactant DIC within the mixed layer (upper 30 m) and were normalized to a salinity of 35. NCP production was calculated between each season from the summer of 2011 to the summer of 2012 (i.e. the change in concentrations between each consecutive season).
according to the equation (Williams, 1993),

\[
NCP = \frac{DIC_{\text{season2}} - DIC_{\text{season1}}}{S_{\text{season2}} - S_{\text{season1}}}\quad \text{(Eq. 3)}
\]

\[= \Delta \text{DIC} \text{ (moles C per unit volume area)}\]

The influx of high-DIC waters (e.g., river discharge) can cause a dampening of the NCP signal. This effect can be accounted for by normalizing DIC to a constant deep-water reference salinity \((S = 35; \text{Millero, 2008})\). Since this equation only reflects the effects of DIC, freshwater influences on alkalinity were accounted for by correction of the seasonal changes in TA (Lee, 2001) using the equation,

\[
\Delta \text{DIC}_{\text{Alk}} = 0.5(\Delta \text{Alk} + \Delta \text{NO}_3^-)\quad \text{(Eq. 4)}
\]

and subtracting this value from the seasonal change in salinity-normalized DIC (nDIC), thus providing an NCP in which the significant process influencing seasonal changes to DIC concentrations is biological productivity (Bates et al, 2005; Mathis et al., 2009; Cross et al., 2012). Error imparted in calculating parameters, including DIC analysis and averaging of nutrient concentrations within the mixed layer, are propagated through our NCP estimates at \(\pm 5\%\) of the final NCP calculation. Error propagated through each NCP estimate is listed with the NCP calculations in Table 1.

### 4.0 Caveats

While seasonal water column DIC concentration changes can be a good approximation to determine seasonal NCP, there are several estuarine processes that we were unable to constrain that likely influenced our NCP estimates and act as additional sources of uncertainty. Some other sources of uncertainty, such as the influence of glacial

Error imparted in calculating parameters, including DIC analysis and averaging of nutrient concentrations within the mixed layer, are propagated through our NCP estimates at \(\pm 5\%\) of the final NCP calculation. Error propagated through each NCP estimate is listed with the NCP calculations in Table 1.

Regional boundaries were selected based on historical and ongoing research in GLBA/Glacier Bay. Due to the sampling schedule and the small areal size of each region, NCP estimates were calculated using data from stations within each region and averaged to provide one regional estimate for each season. Carbon concentrations used for NCP calculations were not analyzed to determine allochthonous vs. autochthonous organic matter origin. Error imparted in calculating parameters, including DIC analysis and averaging of nutrient concentrations within the mixed layer, are propagated through our NCP estimates at \(\pm 5\%\) of the final NCP calculation. Error propagated through each NCP estimate is listed with the NCP calculations in Table 1.

### 4.0 Caveats

While seasonal water column DIC concentration changes can be a good approximation to determine seasonal NCP, there are several estuarine processes that we were unable to constrain that likely influenced our NCP estimates and act as additional sources of uncertainty. Some other sources of uncertainty, such as the influence of glacial
Glacial flour can enhance DIC concentrations in seawater. Therefore, there is the possibility that the inclusion of glacial flour may have increased our DIC concentrations with respect to DIC drawdown from primary production. In this case, our estimates may underestimate NCP. However, we were not able to quantify the amount of glacial flour deposited in Glacier Bay or analyze its composition for this study. In Glacier Bay, the influence of glacial flour is limited to the northern regions (i.e., east and west arms) that are directly influence by glacial outflow, many of which enter the bay along inlets and not the main arms of the bay, possibly reducing the impact of glacial flour at many oceanographic stations in these regions.

Freshwater runoff that enters the bay via glacial streams flows over streambeds and can leach minerals and nutrients from bedrock, enhancing these concentrations in the surface waters of Glacier Bay. While stream water runoff in Glacier Bay was not analyzed for this study, studies of glacial runoff in southeast Alaska have shown allochthonous stream water DOC to be negatively correlated with glacial coverage (Hood, et al., 2009). Examining watersheds along the Gulf of Alaska, Hood et al. (2009) also found that the most heavily glaciated watersheds were a source of the oldest, most labile (66% bioavailable) DOM and that increased input of glacial melt was associated with increased proportions of DOM from microbial sources. As we were unable to chemically analyze glacial runoff in Glacier Bay, our NCP calculations using only changes in DIC concentrations underestimate NCP in the bay, though freshwater input is corrected to some degree by salinity normalized DIC concentrations. The quantification
of freshwater input into the bay is also hindered by the lack of any active gauging stations within the bay (Hill et al., 2009). Some literature suggests that internal waves may form within the lower bay in an area of station 02, known as Sitakaday Narrows. This is an area of constriction with accelerated currents that can produce hydraulic instabilities, potentially causing internal waves that may influence mixing at depth as well as at a distance from this region (Hooge & Hooge, 2002). These internal waves may affect nutrient replenishment to surface waters, as well as mixing of DIC across the mixed layer. This addition of high-DIC waters from depth may also lead to an underestimation of NCP.

5.0 Results

5.1 Spatial and seasonal salinity distributions

Salinity distributions throughout the bay were generally the result of the influence of glacial runoff. During this summer season salinity ranged from 22.9 in surface waters at station 20 to 32.5 in the bottom waters of station 24 in Cross Sound. Isohalines were horizontal down to ~50 m from the upper arms through the upper portion of the lower bay, then became vertical in the lower bay, intersecting the surface just north of station 01 (Fig. 3).

Salinity was more constrained during the fall, with a full water column range between 25.3 in the surface waters at station 07 and 31.4 at depth (~130 m) at station 13. Similar to the previous summer, isohalines remained horizontal from the upper arms to the mid-lower bay near station 01 where they become vertical and intersected the surface.
Salinities in the lower bay near were between ~30 and 31, with the higher salinities at depth in Cross Sound.

During the winter salinity had a narrow range 29.6 and 31.6. The highest salinities were observed in the bottom waters at station 24, though salinity was similar at all depth at this station (~31.4). The lowest salinities (~30) were within the top 10 m of station 12, with similar surface salinities throughout both arms. In the spring, salinity continued to have a narrow range, with bay-wide salinities between ~28.9 at the surface of station 12 and 31.7 in the bottom water of station 24. Salinities below a depth of 50 m were relatively homogenous at ~31 (Fig. 3).

Returning to summer conditions in 2012, a strong salinity gradient was observed in the upper 50 m along the east and west arms. Salinities across the bay ranged from 24.1 in the surface waters of station 12 to 32.2, at depth at station 24. The lowest salinities were observed in the surface waters at the head of both arms, with this low salinity signal stretching south through the through the central bay. Stations within the lower bay had the highest salinities having salinities between ~31 and 32 at all depths.

5.2 Spatial and seasonal distributions of DIC and nitrate

DIC and nitrate are important inorganic components that are consumed during photosynthesis at various rates throughout the year in Glacier Bay. DIC concentrations during the summer of 2011 ranged from ~1400 to 2100 μmol kg⁻¹, with the lowest concentrations in the arms and upper-central bay. Nitrate concentrations throughout the water column ranged from ~2.5 to ~37 μmol kg⁻¹, with slightly less variability in the surface layer (~2.5 and 24 μmol kg⁻¹). Surface nitrate concentrations were low, but...
remained $>$ 5 $\text{umol kg}^{-1}$ at all stations. While there was a large drawdown of nitrate, particularly in spring and summer (as much as 20 $\text{umol kg}^{-1}$ when compared to winter concentrations), surface waters were not depleted at any of the observed stations.

In the fall of 2011, DIC and nitrate concentrations increased in the surface waters, with DIC ranging from ~1700 $\text{umol kg}^{-1}$ to 2040 $\text{umol kg}^{-1}$, while below the surface concentrations reached ~2075 $\text{umol kg}^{-1}$. Water column nitrate concentrations were between ~12 $\text{umol kg}^{-1}$ and 32 $\text{umol kg}^{-1}$ with similar concentrations within surface waters (11 $\text{umol kg}^{-1}$ to 30 $\text{umol kg}^{-1}$) and the lowest concentrations observed in the arms. DIC concentrations were much more constrained during the winter (~1920 $\text{umol kg}^{-1}$ to 2075 $\text{umol kg}^{-1}$) than during previous seasons. Nitrate concentrations ranged from ~12 $\text{umol kg}^{-1}$ to 33 $\text{umol kg}^{-1}$.

During the spring of 2012 DIC and nitrate had reduced concentrations in surface waters across the bay. Surface DIC concentrations were between ~1750 $\text{umol kg}^{-1}$ and 2025 $\text{umol kg}^{-1}$, with water column concentrations reaching ~2075 $\text{umol kg}^{-1}$ (Fig. 4). Nitrate concentrations ranged from ~7 $\text{umol kg}^{-1}$ to ~31 $\text{umol kg}^{-1}$, with an observed surface water maximum of ~20 $\text{umol kg}^{-1}$. Further drawdown of DIC and nitrate in surface waters was observed during the summer of 2012. However, concentrations did not drop as low as was observed during the previous summer. DIC concentrations ranged from ~1545 to 2066 $\text{umol kg}^{-1}$. Nitrate concentrations varied from ~13 to 33 $\text{umol kg}^{-1}$, with surface concentrations between ~17 and 31 $\text{umol kg}^{-1}$. The stations with the lowest DIC and nitrate concentrations were those within the east arm and west arm (Fig. 4).

3 Rates and Masses of NCP
The seasonal transition between the summer and fall of 2011 had the highest rates of NCP observed during the year of study. Rates of NCP were positive in all regions of the bay and were highest within the east and west arms of the bay at 70.3 ± 3.5 and 81.3 ± 4.1 mmol C m⁻² d⁻¹, respectively. A similar NCP rate of 68.9 ± 3.4 mmol C m⁻² d⁻¹ was observed within the lower bay, while the central bay had the lowest rate between of 53.6 ± 2.7 mmol C m⁻² d⁻¹ (Table 1).

Calculated rates of NCP became negative between fall and winter, as well as from winter to spring. Between fall and winter, the lower bay had a rate of -14.2 ± 0.7 mmol C m⁻² d⁻¹ followed by the central bay at -11.5 ± 0.6 mmol C m⁻² d⁻¹. Rates of NCP were negative in the east and west arms (-0.5 ± 0.03 and -1.3 ± 0.1 mmol C m⁻² d⁻¹), respectively. Between the winter and spring of 2012, rates of NCP remained negative within the east and west arms (-36.4 ± 1.8 mmol C m⁻² d⁻¹ and -26.6 ± 1.3 mmol C m⁻² d⁻¹, respectively), and to a lesser degree in central bay (-17.5 ± 0.9 mmol C m⁻² d⁻¹). Positive NCP rate was estimated for the lower bay of 17.6 ± 0.9 mmol C m⁻² d⁻¹. Between the spring and summer of 2012 NCP rates were positive across the bay, with the highest rate in lower bay (19.4 ± 1.0 mmol C m⁻² d⁻¹). The central bay and the east arm had rates of 17.2 ± 0.9 and 15.7 ± 0.8 mmol C m⁻² d⁻¹, respectively, while the west arm had a lower rate at 6.0 ± 0.3 mmol C m⁻² d⁻¹.

The total mass (g C d⁻¹) of carbon produced from NCP was also estimated between each season (Table 1). Production occurred between the summer and fall of 2011, with the greatest production in the lower bay (4.5x10⁷ ± 1.3x10⁷ kg C d⁻¹). The central bay had a large amount of production (2.2x10⁷ ± 1.1x10⁷ kg C d⁻¹), followed by the west and east arms (1.8x10⁷ ± 8.8x10⁶ and 7.6x10⁷ ± 3.8x10⁶ kg C d⁻¹, respectively)...

... [48]
Between the fall and winter the lower bay had carbon production of \(-9.3 \times 10^4 \pm 4.6 \times 10^3\) kg C d\(^{-1}\), while the east arm had a lowest degree of production at \(-5.2 \times 10^4 \pm 2.6 \times 10^3\) kg C d\(^{-1}\). NCP masses in central bay and west arm were also negative (\(-4.7 \times 10^4 \pm 2.3 \times 10^4\) kg C d\(^{-1}\), respectively). Between the winter and spring of 2012 masses in the east and west arms were estimated at \(-3.9 \times 10^4 \pm 2.0 \times 10^4\) kg C d\(^{-1}\) and -\(5.8 \times 10^4 \pm 2.9 \times 10^4\) kg C d\(^{-1}\), respectively while the central bay had a value of \(-7.1 \times 10^4 \pm 3.6 \times 10^4\) kg C d\(^{-1}\). The lower bay was the only region to have a positive NCP of \(1.1 \times 10^4 \pm 5.7 \times 10^3\) kg C d\(^{-1}\).

Transitioning from the spring to summer the lower bay had the greatest production (\(1.3 \times 10^5 \pm 6.3 \times 10^4\) kg C d\(^{-1}\)), followed by the central bay (\(7.0 \times 10^4 \pm 3.5 \times 10^4\) kg C d\(^{-1}\)). The arms exhibited the lowest biomass production, with an NCP in the west arm of \(1.3 \times 10^4 \pm 6.5 \times 10^3\) kg C d\(^{-1}\) and \(1.7 \times 10^4 \pm 8.5 \times 10^3\) kg C d\(^{-1}\) in the east arm.

### 5.4 Spatial and seasonal distribution of POC

During the summer of 2011 surface POC concentrations were between ~12 and ~55 \(\mu\)mol kg\(^{-1}\). Station 20 had the highest POC concentration at all sampled depths (~46 \(\mu\)mol kg\(^{-1}\), ~30, and ~42 \(\mu\)mol kg\(^{-1}\), surface to bottom), while the west arm had the highest POC concentrations below the surface (~33 \(\mu\)mol kg\(^{-1}\) at 50 m and depth). The west and east arms exhibited negative AOU (~80 and ~64 \(\mu\)mol kg\(^{-1}\), respectively). Below the surface concentrations were similar (~9 \(\mu\)mol kg\(^{-1}\)), while surface waters had a POC concentration of ~28 \(\mu\)mol kg\(^{-1}\). Lower bay had relatively lower POC concentrations (~15 \(\mu\)mol kg\(^{-1}\) at all depths).

POC concentrations decreased, especially within surface waters during the fall. A
maximum regional POC concentration (~13 μmol kg⁻¹) was observed in surface waters of the west arm. Below the surface layer POC concentrations were low, between ~5 and ~8 μmol kg⁻¹. A maximum regional surface AOU (~82 μmol kg⁻¹) was estimated for the lower bay and a minimum (~2 μmol kg⁻¹) in the surface waters of the central bay (Fig. 5).

In the winter of 2012 surface water POC concentrations were not found to exceed 20 μmol kg⁻¹ and AOU across the bay were on the order of ~70 μmol kg⁻¹. Surface POC concentrations ranged from ~2 to ~15 μmol kg⁻¹, while POC concentrations at depth varied between ~3 and 16 μmol kg⁻¹. The regional maximum in POC was in the surface waters in the west arm (~11 μmol kg⁻¹). The east arm and lower bay both had maximum POC concentrations in the bottom waters (~14 and ~9 μmol kg⁻¹, respectively).

POC concentration in the surface waters increased during the spring of 2012, primarily within northern regions of the bay. The east arm had the greatest increase in surface POC (~62 μmol kg⁻¹) with concentrations decreasing in the surface water to the south. The west arm and central bay had similar surface POC concentrations of ~35 μmol kg⁻¹, and ~30 μmol kg⁻¹, respectively. The lower bay had the lowest surface POC concentrations with ~13 μmol kg⁻¹, while having the highest rate of NCP and AOU (~93 μmol kg⁻¹). The lower bay subsurface and deepwater AOU values were positive and POC concentrations, ~9 μmol kg⁻¹ each, were the highest among the regions.

AOU values decreased in surface waters across the bay, while rates of NCP were elevated within these waters during the summer of 2012. Surface POC concentrations were highest in the east arm (~50 μmol kg⁻¹), while below the surface layer, POC concentrations decreased, ranging from ~4.5 to ~7 μmol kg⁻¹ at 50 m and ~5 to ~8 μmol kg⁻¹ at depth. The west arm and central bay regions had surface POC concentrations of...
~23 μmol kg⁻¹ and the lower bay exhibited the lowest surface POC concentration with
~13 μmol kg⁻¹.

5.5 Relationship between DIC and DO

During the summer of 2011, DO concentrations ranged from ~190 to ~400 μmol kg⁻¹. All samples below the surface layer, as well as surface samples within the lower bay followed the Redfield ratio, with concentrations at depth between ~190 and 280 μmol kg⁻¹ (Fig. 6). Surface samples of stations within the arms and central bay had high DO concentrations and low DIC. Surface DO was higher than that at depth, ranging between ~230 and 400 μmol kg⁻¹. However, in the lower bay DIC concentrations remained elevated (~2030 μmol kg⁻¹) and DO concentrations were low (~240 μmol kg⁻¹). During the fall, surface samples within the arms and central bay continued to deviate from Redfield. Surface DO concentrations ranged from ~210 to ~330 μmol kg⁻¹ and corresponded with reduced surface DIC concentrations. At depth, DO concentrations varied between ~200 and 280 μmol kg⁻¹ with C:O ratios close to Redfield.

All samples, at the surface and at depth, followed Redfield closely with surface waters having slightly higher DO and lower DIC concentrations than those at depth during the winter of 2012. Surface water DO concentrations were between 250 and ~280 μmol kg⁻¹, while deeper waters ranged from ~230 to 255 μmol kg⁻¹.

In the spring, DIC was drawn down and DO concentrations increased, having a range between ~270 and 410 μmol kg⁻¹. DO concentrations were amplified while DIC was reduced at stations in the northern-most regions of both arms. These samples deviated the most from Redfield, while the remaining samples adhered to the Redfield.
ratio. Below the surface layer, DO concentration throughout the bay ranged from ~250 to 280 μmol kg\(^{-1}\). During the summer of 2012, the surface waters within the two arms and central bay continued to diverge from Redfield. DIC concentrations within the more northern regions of the bay (east arm, west arm, and central bay) were increasingly drawn down, while DO concentrations remained elevated. Surface DO concentrations ranged from ~260 to ~410 μmol kg\(^{-1}\), with lower DO concentrations at depth, varying from 200 - ~270 μmol kg\(^{-1}\).

### 5.6 Air-Sea gas flux

During the summer of 2011 winds were relatively low, at ~1.6 m s\(^{-1}\), with surface waters of the central bay and the west arm were undersaturated with respect to atmospheric CO\(_2\) with \(p\text{CO}_2\) values of ~250 μatms. The central bay and the west arm acted as minor sinks (~0.3 ± 0.02 mmol C m\(^{-2}\) d\(^{-1}\) each). The lower bay and east arm had much higher seawater \(p\text{CO}_2\) values of ~488 μatms and ~463 μatms and acted as sources for atmospheric CO\(_2\) of ~0.2 ± 0.01 mmol C m\(^{-2}\) d\(^{-1}\) for each region (Fig. 7).

During the fall of 2011, winds increased slightly to ~2.0 m s\(^{-1}\) and surface waters in all regions of the bay were oversaturated with respect to the atmospheric CO\(_2\). The lower bay experienced the highest \(p\text{CO}_2\) at ~670 μatms and acted as the largest source for atmospheric CO\(_2\) with a flux of ~1.1 ± 0.06 mmol C m\(^{-2}\) d\(^{-1}\). The central bay also had elevated \(p\text{CO}_2\) with ~510 μatms leading to outgassing of ~0.5 ± 0.03 mmol C m\(^{-2}\) d\(^{-1}\). The east arm had a \(p\text{CO}_2\) and flux values similar to that of the central bay (\(p\text{CO}_2\) ~514 μatms; flux = ~0.5 mmol ± 0.03 C m\(^{-2}\) d\(^{-1}\)). Air-sea CO\(_2\) flux in the west arm was ~0.3 ±
Surface waters during the winter of 2012 were oversaturated in CO$_2$ with respect to the atmosphere and all regions experienced outgassing, with average wind speeds of ~2.1 m s$^{-1}$. Regional pCO$_2$ values were more constrained, especially within the arms and central bay, ranging from ~400 μatms in the west arm and central bay to ~432 μatms in the east arm.

Similar pCO$_2$ values and seawater temperatures (~3.5°C), led the west arm and central bay to experience comparable CO$_2$ fluxes of ~0.03 ± 0.002 and 0.06 ± 0.003 mmol C m$^{-2}$ d$^{-1}$. The east arm had a slightly higher surface temperature (~4.1°C) and flux, with ~0.18 ± 0.01 mmol C m$^{-2}$ d$^{-1}$, while the lower bay had a slightly higher CO$_2$ flux of ~0.76 ± 0.04 mmol C m$^{-2}$ d$^{-1}$.

In the spring, seawater temperatures increased slightly to ~5°C across the bay while salinity remained similar to winter values (~29 to 31). However, all regions except the lower bay transitioned to sinks for atmospheric CO$_2$. pCO$_2$ in the lower bay remained oversaturated with respect to CO$_2$ at ~423 μatms and had a flux of ~0.11 ± 0.01 mmol C m$^{-2}$ d$^{-1}$. Within the other three regions of the bay, surface water temperatures increased by just over 1°C. However, pCO$_2$ decreased in the surface waters and these regions acted as sinks for atmospheric CO$_2$. The east arm had the greatest decrease in pCO$_2$, dropping from ~432 μatms to ~167 μatms and exhibiting seasonal outgassing of ~0.87 ± 0.04 mmol C m$^{-2}$ d$^{-1}$. The central bay and west arm regions were also seasonal sinks for CO$_2$, taking up ~0.39 ± 0.02 mmol C m$^{-2}$ d$^{-1}$ in the central bay and ~0.60 ± 0.03 mmol C m$^{-2}$ d$^{-1}$ in the west arm.
During the summer of 2012 $p$CO$_2$ in the east arm increased to ~337 μatms with a flux of ~0.13 ± 0.01 mmol C m$^{-2}$ d$^{-1}$ of ingassing. The central bay had a $p$CO$_2$ of ~200 μatms and a flux of ~0.44 ± 0.02 mmol C m$^{-2}$ d$^{-1}$. The lower bay and west arm, acted as sources for atmospheric CO$_2$, having $p$CO$_2$ values of ~411 μatms and ~507 μatms, respectively, while the lower bay experienced a near-neutral flux of ~0.04 ± 0.002 mmol C m$^{-2}$ d$^{-1}$. The west arm was oversaturated with respect to atmospheric CO$_2$ with a $p$CO$_2$ of ~507 μatms and a flux of ~0.26 ± 0.01 mmol C m$^{-2}$ d$^{-1}$.

6.0 Discussion

6.1 Relationships of DIC, Nitrate, and Dissolved Oxygen

DIC, nitrate and DO are important indicators of biological production in a marine ecosystem. One way they can be used as biological production indicators is through Redfield ratios. Carbon and oxygen have a C:O Redfield ratio of 106:170 (Anderson et al., 1994) and the carbon to nitrate Redfield ratio is 106:16.

During the summer of 2011 variability in DIC, nitrate and dissolved oxygen concentrations within the surface waters were a result of primary production, dilution from glacial discharge, or a combination of both processes. Surface waters in the arms and upper-central bay deviated from Redfield ratios for C:O and C:N (Figs. 6 and 8). Waters below this surface layer followed the Redfield ratios throughout the year. Nitrate and phosphate concentrations in the surface waters were not observed to reach depletion during the summer, indicating that they were being continuously supplied to the surface layer and that phosphate (data not shown) was not limiting. Sustained nutrient concentrations and nutrient replenishment may be the result of several physical interactions within the bay, including wind, tidal and internal wave mixing, especially...
over shallow sills at the mouth of the bay and at the entrance to the east arm.

Increases in DO and the reduction in macronutrient concentrations, including DIC, within the more northern arms of the bay was due to primary production coupled with the influence of glacier runoff and salinity-driven stratification limiting mixing and nutrient replenishment in the mixed layer. In the fall of 2011, DIC and nitrate concentrations increased while DO decreased in the surface waters as primary production slowed and wind mixing increased. Due to decreasing primary production nutrient concentrations were similar within surface waters with the lowest concentrations observed in the arms where glacial runoff was still impacting surface waters. Surface water ratios for C:O and C:N deviated from the Redfield ratios, but less so than observed during summer as primary production began to decrease during the fall (Figs. 6 and 8).

During the winter of 2012, increased wind mixing and the reduction of glacial input led to deeper water column mixing, with much more constrained DIC and nitrate concentrations. During the winter nitrate and DIC concentrations continued to increase, with C:O and C:N Redfield ratios indicated a decrease in primary production and increase in mixing (Figs. 6 and 8). While DIC and nitrate concentrations fell near the Redfield ratio, they deviated slightly from Redfield at the highest nitrate concentrations (Fig. 4). This may have been due to nitrification of ammonium by bacteria leading to an increase the nitrate concentration. Another possibility is ‘carbon overconsumption’, the process in which more DIC is taken up than that inferred from the C:N Redfield ratio (Voss et al., 2011). Explanations for carbon overconsumption include the preferential remineralization of organic nitrogen (Thomas and Schneider, 1999) or an increased release of dissolved organic carbon (Engel, et al., 2002; Schartau et al., 2007).
As temperatures began to warm in the spring of 2012, the onset of glacial melt and primary production reduced DIC and nitrate, while increasing DO concentrations in surface waters across the bay. DIC and nitrate correlated closely with the Redfield ratio except for two surface samples located at the northernmost ends of each arm (Fig. 8). This deviation may be explained by the fact that these stations were the first to be influenced by glacial runoff during the onset of the glacial melt season.

Further reduction in DIC and nitrate concentrations in surface waters was observed during the summer of 2012 as primary production intensified, increasing DO concentrations. Low nutrient glacial runoff was highest at this time of year, affecting surface water DIC and nitrate concentrations within the arms. However, concentrations did not drop as low as was observed during the previous summer. Macronutrients did not reach depletion during the summer of 2012, implying they were not the limiting primary productivity, possibly due to nutrient replenishment via tidal pumping. Surface nitrate concentration continued to deviate from the C:N Redfield ratio as these macronutrients were increasingly drawn down by primary productivity and diluted by glacier runoff (Fig. 8). Surface waters in several regions also deviated from the C:O Redfield ratio (Fig. 6). The stations most affected were those within the east arm and west arm, as well as upper central bay, where freshwater influence was greatest. Mixing of nutrient-rich marine waters from the Gulf of Alaska likely offset much of the drawdown from primary production and allowed these surface waters within the lower bay to fall closer to the Redfield ratio.

6.2 NCP
The seasonal transition between the summer and fall of 2011 had the largest rates of NCP observed during the year of study. During this time all NCP rates were positive, signifying enhanced primary productivity in the mixed layer. Rates of NCP became negative during the seasonal transitions from fall to winter, as well as from winter to spring. These negative NCP values indicate that air-sea fluxes (discussed in Section 5.6) and organic matter respiration were prominent, increasing CO$_2$ (DIC) concentrations in the surface waters and overwhelming any weaker signal from primary production. Between the fall and winter, the lower bay experienced the highest degree of CO$_2$ flux when compared to biological production. The biological production was overwhelmed by CO$_2$ influx in the east and west arms, but to a less degree than in regions to the south. Between the winter and spring of 2012 the lower bay was the only region where biological production dominated the CO$_2$ flux with a positive NCP rate, reflecting the region’s nutrient-rich marine influence from the Gulf of Alaska. The CO$_2$ flux signal exceeded NCP within the east and west arms of the bay and, to a lesser extent, the central bay. Transition from the spring to summer of 2012, primary production was evident in the NCP rates. The west arm experienced a lower rate of NCP, possibly the result of the strong low-macronutrient glacial influences along the arm, which may work to hinder production. Additionally, large volumes of glacial flour imparted into the surface waters from runoff during summer may have limited the photic depth and thus impeded some productivity in the upper arms of the bay. The total mass of carbon produced between seasons via NCP was also estimated (Table 1). Between the summer and fall of 2011, we observed the greatest production of organic carbon of any seasonal transition, with the largest production signal in the lower
bay and decreasing to the north as glacial influence increased. Elevated production
estimates within the lower could be due to continued nutrient replenishment to surface
waters as a result of mixing with the more marine waters outside of the bay.

Despite all regions of the bay being dominated by air-sea CO₂ flux during
between the fall and winter seasons (Table 1), there was a substantial contrast in
magnitudes of estimates between the marine-dominated lower bay and the glacially-
influenced east arm. These differences in magnitude were likely the result of a higher
degree of wind and tidal mixing at stations outside of and near the mouth of the bay,
allowing this region to have elevated air-sea flux when compared to the east and west
arms (Fig. 7).

The production signal within the arms and central regions of the bay continued to
be overwhelmed by air-sea flux between the winter and spring of 2012 (Table 1). While
production estimates remained negative in the northern regions of the bay, the lower bay
had a positive NCP mass signifying increased primary production and a decrease in air-
sea flux in this region. This increase in NCP in the lower bay may be the result of
earlier nutrient replenishment via the more marine waters outside of the bay. Between the
spring and summer there was increased production across the bay as stratification
strengthen and the hours of daylight increased, with the largest production estimates in
the lower bay. The east and west arms exhibited the lowest biomass production, likely
hindered by the inundation of low-nutrient glacial runoff that formed a fresh surface layer
and imparted glacial flour into the surface waters in these regions.

6.3 Air-Sea Flux
Aside from primary production, air-sea carbon dioxide (CO$_2$) flux also impacts carbon concentrations within surface waters. In Glacier Bay, air–sea fluxes varied regionally and seasonally between the summer of 2011 and the summer of 2012. During the summer of 2011 winds were relatively low, reducing turbulent mixing, allowing for stratification and, thus, primary production. Surface waters in the lower bay and east arm acted as sources for atmospheric CO$_2$, while the central bay and the west arm acted as sinks (Fig. 7). Drawdown of CO$_2$ in the west arm may be attributed to primary production, as well as the influx of low nutrient glacial melt. The central bay has been noted to have elevated production levels (Hooge and Hooge, 2002) that may account for the drawdown of DIC and the region’s sink status. Within the east arm seawater temperatures were high, increasing the pCO$_2$ of these waters and, combined with influence of the reduced TA concentrations, resulted in an oversaturation of CO$_2$ in the seawater with respect to the atmosphere, overwhelming any effect from DIC drawdown via primary production and making this region a source for atmospheric CO$_2$. Turbulent mixing across and outside the sill, as well as through Sitakaday Narrows, likely reduced stratification and enhanced air-sea flux, causing this region to be a source for atmospheric CO$_2$. In the fall of 2011, winds increased slightly and all surface waters across the bay experienced oversaturation with respect to the atmospheric CO$_2$, with the lower bay acting as the strongest regional source (Fig. 7). The high pCO$_2$ values observed during fall, despite strong DIC drawdown during summer, may be the result of a variety of interactions. Reduced glacial runoff during fall increased TA concentrations (Reisdorph and Mathis, 2014) and surface water temperatures declined allowing them to hold more
CO₂ while mixing brought DIC-rich waters from depth to the surface. Increased winds also likely led to enhanced turbulent mixing across the bay. During the winter of 2012 surface waters across all regions of the bay continued to experience outgassing (Fig. 7), though to a lesser degree than during fall. The lower bay experienced the largest degree of outgassing, likely due to its more turbulent mixing than other regions. Despite winter having the lowest seawater temperatures, wind mixing peaked and likely allowed for CO₂-rich waters from depth and the air to enter the surface waters, increasing pCO₂ in all regions of the bay.

Several regions of Glacier Bay transitioned to sinks for atmospheric CO₂ during the spring of 2012 as primary production increased and winds slowed. The lower bay was the exception, remaining oversaturated with respect to CO₂ and continuing to act as a minor source for atmospheric CO₂. In the more northern regions, surface waters experienced a slight increase in surface temperatures, but due to the onset of spring productivity DIC was drawn down in the surface waters, decreasing the pCO₂ and allowing them to become sinks for atmospheric CO₂. The east arm experienced the largest decrease in pCO₂ and became the largest sink region within the bay, while the west arm and central bay underwent similar flux transitions as primary production increased, drawing down DIC in the surface waters. Within the arms, the onset of glacial melt may have aided in setting up stratification, also helping to lead to larger sink statuses within these regions.

During the summer of 2012, waters in the northern regions becoming increasingly saturated with respect to atmospheric CO₂. While, pCO₂ in the east arm did increase from spring values, perhaps due to a small increase in surface water temperatures and reduced
in TA from glacial runoff, it was still undersaturated with respect to atmospheric $pCO_2$. Atmospheric $CO_2$ uptake within the central bay strengthened slightly from spring as $pCO_2$ in this region decreased, likely due to high levels of primary production in this region, as well as high nutrient replenishment from tidal mixing between the waters of lower bay and the stratified waters within the central bay (Hooge & Hooge, 2002).

Conversely, the lower bay remained a minimal source for atmospheric $CO_2$, while the west arm transitioned into source during the summer. The lower bay experiences the highest degree of turbulent or tidal mixing across the sill, within Cross Sounds, and through Sitakaday Narrows, inhibiting stratification and primary production and causing it act as a source for atmospheric $CO_2$ year-round. The difference in the sink/source status of the east and west arms of the bay was likely the result of differences in glacial influences, with the west arm more influenced by low-TA glacial runoff as it has the majority of the tidewater glaciers along its length. These glaciers caused a higher degree of TA and DIC dilution than was observed within the west arm.

7.0 Conclusions

Glacier Bay experiences a high degree of spatial and temporal throughout the year. Environmental influences vary seasonally along a gradient from the glacially-influenced northern regions within the arms to the marine-influenced lower bay. This imparts spatial differences in stratification and macronutrient availability that effect biological processes and thus, rates of NCP within each of the four pre-defined regions of the Glacier Bay.

Despite Glacier Bay’s limited exchange with the marine waters of the Gulf of...
Alaska, it has been observed to support elevated primary production through most of the year (Hooge & Hooge, 2002), perhaps due to tidal pumping. However, rapid deglaciation within Glacier Bay has imparted a high volume of fresh glacial runoff, a portion of which has been from tidewater glaciers that melt directly into the bay, affecting stratification, macronutrient concentrations and influencing air-sea CO₂ exchange.

Rates of NCP were positive across the bay between the summer and fall of 2011, as well as between the spring and summer of 2012 during peak times of primary production. NCP was highest during the transition between summer and fall of 2011, with regional NCP rates ranging from ~54 to ~80 mmol C m⁻² d⁻¹. Rates during the summer of 2012 were lower, between ~6 and ~20 mmol C m⁻² d⁻¹.

Between the fall of 2011 and winter of 2012, as well as between the winter and spring of 2012, air-sea gas exchange overwhelmed any production signal across the bay, especially during the fall (Fig. 7; Table 1). The one exception was lower bay between winter and spring where NCP rates were positive, likely due to earlier replenishment of nutrients from marine waters outside the bay.

The impact of rapid deglaciation in Glacier Bay can be observed in the seasonal impacts on the carbon cycling and NCP in this estuarine system. This study enhances the limited biogeochemical literature regarding Glacier Bay and includes one of the more robust datasets from Glacier Bay. The influence of surrounding glaciers, especially tidewater glaciers, has the potential to significantly impact the efficiency and makeup of the marine food web within Glacier Bay in unknown ways with unknown consequences. Better understanding of the influences of NCP can help identify possible these outcomes.
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Fig. 1: Glacier Bay location and oceanographic sampling station map - Blue lines denote regional boundaries. Red dots show all oceanographic station locations with station number. Blue stars represent ‘core’ station location. Lower bay, central bay, east, west arm.
Figure 2: Bathymetry of Glacier Bay – Bathymetric map of Glacier Bay
Figure 3: Seasonal distribution of salinity. Spatial and seasonal distribution of salinity in the water column.
Figure 4: Spatial distribution of DIC and nitrate. Spatial and seasonal distribution of DIC in the water column. Contours represent nitrate concentrations.
<table>
<thead>
<tr>
<th>Seasonal transition</th>
<th>Region</th>
<th>Regional Area (m²)</th>
<th>NCP rate (mmol C m⁻² d⁻¹)</th>
<th>NCP mass (kg C d⁻¹)</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Summer and Fall</strong></td>
<td>Lower Bay</td>
<td>5.44x10⁸</td>
<td>68.9 ± 3.5</td>
<td>4.5x10⁴ ± 2.3x10⁴</td>
</tr>
<tr>
<td></td>
<td>Central Bay</td>
<td>3.40x10⁸</td>
<td>53.6 ± 2.7</td>
<td>2.2x10⁴ ± 1.1x10⁴</td>
</tr>
<tr>
<td></td>
<td>West Arm</td>
<td>1.80x10⁸</td>
<td>81.3 ± 4.1</td>
<td>1.8x10⁴ ± 8.8x10³</td>
</tr>
<tr>
<td></td>
<td>East Arm</td>
<td>9.00x10⁷</td>
<td>70.3 ± 3.5</td>
<td>7.6x10³ ± 3.8x10³</td>
</tr>
<tr>
<td><strong>Fall and Winter</strong></td>
<td>Lower Bay</td>
<td>5.44x10⁸</td>
<td>-14.2 ± 0.7</td>
<td>-9.3x10³ ± 4.6x10³</td>
</tr>
<tr>
<td></td>
<td>Central Bay</td>
<td>3.40x10⁸</td>
<td>-11.5 ± 0.6</td>
<td>-4.7x10³ ± 2.3x10³</td>
</tr>
<tr>
<td></td>
<td>West Arm</td>
<td>1.80x10⁸</td>
<td>-1.3 ± 0.1</td>
<td>-2.7x10³ ± 135.7</td>
</tr>
<tr>
<td></td>
<td>East Arm</td>
<td>9.00x10⁷</td>
<td>-0.5 ± 0.0</td>
<td>-515.7 ± 25.8</td>
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<tr>
<td><strong>Winter and Spring</strong></td>
<td>Lower Bay</td>
<td>5.44x10⁸</td>
<td>17.6 ± 0.9</td>
<td>1.1x10⁵ ± 5.7x10³</td>
</tr>
<tr>
<td></td>
<td>Central Bay</td>
<td>3.40x10⁸</td>
<td>-17.5 ± 0.9</td>
<td>-7.1x10⁴ ± 3.6x10³</td>
</tr>
<tr>
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<td>West Arm</td>
<td>1.80x10⁸</td>
<td>-26.6 ± 1.3</td>
<td>-5.7x10⁴ ± 2.9x10³</td>
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<tr>
<td></td>
<td>East Arm</td>
<td>9.00x10⁷</td>
<td>-36.4 ± 1.8</td>
<td>-3.9x10⁴ ± 2.0x10³</td>
</tr>
<tr>
<td><strong>Spring and Summer</strong></td>
<td>Lower Bay</td>
<td>5.44x10⁸</td>
<td>19.4 ± 1.0</td>
<td>1.3x10⁵ ± 6.3x10³</td>
</tr>
<tr>
<td></td>
<td>Central Bay</td>
<td>3.40x10⁸</td>
<td>17.2 ± 0.9</td>
<td>7.0x10⁴ ± 3.5x10³</td>
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<td>9.00x10⁷</td>
<td>15.7 ± 0.8</td>
<td>1.7x10⁴ ± 846.9</td>
</tr>
</tbody>
</table>

Table 1: Regional rates and masses of NCP—NCP by region in Glacier Bay based on the change in salinity-normalized DIC concentrations between seasons.
Fig. 5: Seasonal POC vs. depth vs. AOU - Seasonal scatter plots of POC concentrations vs. depth for each season between the summer of 2011 through the summer of 2012.

Color bar represents AOU in $\mu$mol kg$^{-1}$.
Fig. 6: Seasonal DIC vs. DO vs. depth - Scatter plots of DIC concentrations vs. DO concentrations for each season between the summer of 2011 and the summer of 2012. Color bar represents depth in m. The red line depicts the C:O Redfield ratio of 106:170. Dotted circles highlight samples that deviate from Redfield.
Fig. 7: Air-sea CO$_2$ flux – Seasonal air-sea CO$_2$ fluxes by region in mmoles C m$^{-2}$ d$^{-1}$. Blue represents the summer of 2011, red = fall of 2011, green = winter of 2012, purple = spring of 2012, yellow = summer of 2012.
Fig. 8: Seasonal DIC vs. NO$_3^-$ vs. depth - Scatter plots of DIC concentrations vs. NO$_3^-$ concentrations for each season between the summer of 2011 and the summer of 2012. Color bar represents depth in m. The red line depicts the C:N Redfield ratio of 106:16. Dotted circles highlight samples that deviate from Redfield.