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# The metabolic response of pteropods to ocean acidification reflects natural CO<sub>2</sub>-exposure in oxygen minimum zones

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# Abstract

Shelled pteropods (Thecosomata) are a group of holoplanktonic mollusks that are believed to be especially sensitive to ocean acidification because their aragonitic shells are highly soluble. Despite this concern, there is very little known about the physio-

- Iogical response of these animals to conditions of elevated carbon dioxide. This study examines the oxygen consumption and ammonia excretion of five pteropod species, collected from tropical regions of the Pacific Ocean, to elevated levels of carbon dioxide (0.10%, 1000 ppm). Our results show that pteropods that naturally migrate into oxygen minimum zones, such as *Hyalocylis striata*, *Clio pyramidata*, *Cavolinia longirostris* and the provide the string of the provide the provide the string of the provide the string of the provide the pr
- *Creseis virgula*, were not affected by carbon dioxide at the levels and duration tested. *Diacria quadridentata*, which does not migrate, responds to high carbon dioxide conditions with reduced oxygen consumption and ammonia excretion. This indicates that the natural chemical environment of individual species influences their resilience to ocean acidification.

# 15 1 Introduction

Marine systems are a significant sink for the excess carbon produced by human activities. Since pre-industrial times, atmospheric carbon dioxide ( $CO_2$ ) levels have risen from 280 ppm to the current 390 ppm (Feely et al., 2004). This rise in atmospheric  $CO_2$  concentration has grown at a slower rate than human output, a discrepancy which

- is due to the buffering capacity of the Earth's marine system. About 30 % of anthropogenic  $CO_2$  ends up in the surface waters of the ocean (Feely et al., 2009). As this gas dissolves and interacts with seawater, it dissociates into bicarbonate and free hydrogen ions, a process that reduces the ocean's pH and carbonate ion concentration. The pH of the ocean has already dropped by ~ 0.1 units relative to preindustrial levels
- <sup>25</sup> and is predicted to drop another 0.2 to 0.3 in the next one hundred years (Haugan and Drange, 1996; Caldeira and Wickett, 2003, 2005). Acidification has been identified as

the third most pervasive human impact on the ocean (Halpern, 2008). It is therefore important to understanding the physiological response of marine organisms to elevated  $CO_2$  and reduced pH.

Elevated seawater CO2 can be detrimental to organisms because it crosses biologi-

- cal membranes and reacts with intra- and extracellular fluids just as it does with ocean waters. The resulting internal acidosis influences a number of physiological processes (Seibel and Walsh, 2001; Seibel and Fabry, 2003). Organisms have some capacity to compensate for the pH change but at an energetic cost that may result in physiological trade-offs (Wood et al., 2008). The capacity of organisms to control acid-base balance
- is dependent on the rate of metabolic CO<sub>2</sub> production as well as exposure to natural environmental CO<sub>2</sub> levels. As a result, each species has a different tolerance level for environmental pH changes, with implications for growth, fecundity and survival. Ocean acidification is also of particular concern for shell bearing organisms since
- calcification requires additional energy in the face of decreased carbonate ion concentration in seawater (Cohen and Holcomb, 2009). Anthropogenic  $CO_2$  has already reduced the saturation state of calcium carbonate in the tropics. As a result, calcite precipitation has dropped by 6–11%, and, based on climate models, this could reach 35% in the next 100 yr (Kleypas et al., 1999). The cosomatous pteropods have received a great deal of attention in ocean acidification discussions because they produce thin
- shells made of aragonite, a highly-soluble form of calcium carbonate. These pelagic gastropods are found throughout the world, predominantly in near-surface seawater, although deep-sea species are also known to exist (Lalli and Gilmer, 1989). Studies of the potential effects of ocean acidification on pteropods have primarily focused on polar species because of their abundance and importance in regional food-webs and
- <sup>25</sup> carbon biogeochemical cycles (Pakhomov et al., 2002; Accornero et al., 2003; Armstrong et al., 2005; Manno et al., 2010) and because the polar oceans are expected to reach undersaturation first due to the increased solubility of CO<sub>2</sub> in cold water. The Arctic species *Limacina helicina* shows a 28% decrease in calcification at 780 ppm CO<sub>2</sub>, although it is capable of precipitating aragonite at low saturation states (Comeau

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et al., 2009, 2010a). Juveniles of *Limacina helicina* respond to elevated (780 and 1100 ppm  $CO_2$ ) with changes in shell diameter, shell increment and shell degradation (Lischka et al., 2011). The impact of  $CO_2$  on the metabolic rate of this species, and its congener *Limacina helicina antarctica*, are less clear. It appears that environmental stressors, such as temperature and food availability, produce synergistic responses to  $CO_2$ , either increasing or decreasing metabolism (Comeau 2010a, b; Seibel et al.,

2011). Little is known about the physiology of tropical pteropod species, and the impact of hypercapnia (high  $CO_2$ ) on metabolism has been reported only for one warm water species. *Carelinia inflava* (Company et al., 2010a). There are no studies which relate

- <sup>10</sup> species, *Cavolinia inflexa* (Comeau et al., 2010a). There are no studies which relate the distribution of pteropods to regions of naturally occurring high carbon dioxide concentrations. In some pelagic ecosystems, such as the Eastern Tropical Pacific (ETP) and the Gulf of California, a pronounced oxygen minimum zone (OMZ) exists in which respiration below the photic layer outpaces mixing to create a region of low oxygen
- and elevated carbon dioxide. Specifically, in the Pacific around depths of 200 m, CO<sub>2</sub> levels reach approximately 1000 ppm and 400 ppm at 10° N and 30° N, respectively (Fabry et al., 2008). These pHs correspond to a decreasing saturation state of calcium carbonate. Waters are undersaturated with respect to aragonite ( $\Omega_a < 1$ ) at 1000 ppm CO<sub>2</sub>, suggesting that there would be passive dissolution of pteropod shells near 10° N.
- <sup>20</sup> Testing whether these zones of low pH (< 7.6) act as a barrier to pteropod distribution, and investigating the impact of hypercapnia on the metabolism of resident species will provide insight to the potential effects of anthropogenic acidification on organisms living in tropical surface waters. If pteropods are naturally found at hypercapnic conditions, it is unknown whether they will respond to laboratory exposure to CO<sub>2</sub> with an increase
- in metabolism as some metabolic processes become up-regulated to deal with acidification, with a lowered metabolic rate to withstand the energy limitation brought on by acidosis or whether they will be unaffected. Here, we report the effect of short periods of hypercapnia (6–18 h, 1000 ppm CO<sub>2</sub>) on the routine metabolic rate (oxygen consumption) and ammonia excretion of five species of the cosomatous pteropods. We

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then compare these organismal responses with the natural distribution of these species in relation to the OMZ of the ETP.

# 2 Methods

We collected individuals representing five species of the cosomatous pteropods (Hyalo-

- cylis striata, Clio pyramidata, Cavolinia longirostris, Creseis virgula and Diacria quadridentata) from three sites in the Pacific Ocean between June 2007 and January 2009 (Table 1). Hydrographic profiles of all three regions were assembled from CTD casts made during the time of organism collection (Fig. 1). Profiles of pH were calculated using a Multi-parameter Inorganic Carbon Analyzer (MICA) in the ETP in 2008 (Byrne and
- Elliott, unpublished data). We retrieved animals from a 61 cm-diameter 335 μm-mesh bongo net trawl, a 10 m<sup>2</sup> Tucker trawl with a thermally protected cod end (Childress et al., 1978), or using SCUBA (Haddock and Heine, 2005).

After capture, organisms were put into 0.2 micron-filtered water at densities < 10 individuals  $I^{-1}$  and left to acclimate at 20  $^\circ C$  for at least eight hours. During experi-

- <sup>15</sup> ments, we put animals into glass syringe respiration chambers with a known volume (10–50 ml) of 0.2 micron-filtered seawater for no less than six hours. The water contained 25 mg of Streptomycin and 25 mg of Ampicillin I<sup>-1</sup> and had been bubbled with certified gas to achieve CO<sub>2</sub> concentrations of ~ 0.10 % (1000 ppm) or allowed to remain at standard air saturation (0.03 %, 380 ppm). To measure the pH of the water
- used in the hypercapnic studies, we used a flow-through water-jacketed pH electrode (Microelectrodes, Bedford NH, #16-705). The pH of hypercapnic treatments averaged 7.96 ± 0.10, whereas normocapnic water averaged pH 8.29 ± 0.04. At the beginning of each experiment, we set up a blank syringe to monitor background respiration of microbes using identically bubbled water.
- At the end of each respiration incubation (6–18 h), an aliquot of water was withdrawn from both the experimental and the blank chambers using a 500  $\mu$ l airtight Hamilton syringe and injected past a Clarke-type O<sub>2</sub> electrode (Strathkelvin Instruments, North

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Lanarkshire, UK) in a water-jacketed injection port (Marsh and Manahan, 1999). Experimental values were subtracted from blank values and the resulting computed  $O_2$  consumption rates are reported in  $\mu$ mol g<sup>-1</sup> h<sup>-1</sup> (wet mass). A second sample of water was immediately dawn and frozen in a cryovial at -80 °C. These samples were later thawed

- and NH<sub>3</sub> concentration (μmol g<sup>-1</sup> h<sup>-1</sup> wet mass) was measured using the indophenol blue colorimetric assay (Ivancic and Degobbis, 1984). Organisms were weighed using a motion-compensated shipboard balance system (Childress and Mickel, 1980), then frozen in liquid nitrogen. Upon return to shore, a subset of animals were reweighed using a pinnacle series analytical balance (± 0.001 g, Denver Instruments) to verify
- the accuracy of the field measurements. Statistical analyses were conducted using the STATISTICA software package (StatSoft). Tests were reported as significant if p < 0.05.

In the ETP, diel vertical distribution of zooplankton was sampled using a vertically stratified MOCNESS (Multiple Opening/Closing Net and Environmental Sensing Sys-

- tem; Wiebe et al., 1976; Wishner component of the ETP program). Samples were collected from 0–400 m using 153-µm mesh nets in sampling intervals which varied from 10 m to 150 m thick during the day and night (Table 2). These samples were split using a flat-bottomed Motoda splitter and were preserved in a 4 % sodium borate-buffered formalin and sea water solution. Upon return to the laboratory, we separated each
- sample by size-fraction using a 64 µm-mesh sieve. Pteropods were picked out from this subsample and identified to species using a dissecting microscope. For this paper, the presence or absence of individual species of pteropods was documented for each net and assembled into a vertical profile to provide a diel vertical pattern of species specific distribution. Quantitative abundances are reported in Maas et al. (2011).

### 25 3 Results

The  $O_2$  consumption of pteropods was impacted by organismal mass (Fig. 2, Table 3). Using a one-way ANCOVA to account for the effect of mass (continuous predictor), we

found that only *Diacria quadridentata* had a significant difference in the average rate of  $O_2$  consumption between hypercapnic and normocapnic treatments (groups; Tables 4 and 5; Fig. 3). When exposed to elevated  $CO_2$ , *Diacria quadridentata* responded with a significant depression in  $O_2$  consumption rate (~53%, p = 0.033). Similarly, *Diacria* 

 $_{5}$  quadridentata was the only species to respond to hypercapnic conditions with a significant reduction in NH<sub>3</sub> excretion (~ 63 %, p = 0.009, Tables 4 and 5; Fig. 4). Ratios of O<sub>2</sub> consumption and NH<sub>3</sub> excretion (O:N) were not statistically different between treatments for any of the species studied (Tables 4 and 5).

MOCNESS sampling revealed distinct differences in vertical distribution both between species and between stations (Fig. 5). *Diacria quadridentata* was the only completely non-migratory species; it was always found above the mixed layer (~ 30 m). The distribution of all other pteropods included depths below the mixed layer into the low O<sub>2</sub> water at the Costa Rica Dome. Generally organisms were found at depth during the daytime and nearer the mixed layer during the evening, although portions of the *Hyalo*-

- cylis striata, Cavolinia longirostris and Creseis virgula populations were found at depth during the night. At the Tehuantepec Bowl, where the transition to the OMZ was generally more abrupt and severe, patterns of distribution were quite different. In this region we never collected *Clio pyramidata*, and *Creseis virgula* was present only at < 100 m, unlike their distribution to 350–400 m during both the day and night at the Costa Rica</p>
- Dome. Only *Hyalocylis striata* was found at similar depths at both stations. These distributions reveal that most pteropod species in the ETP daily inhabit regions of low O<sub>2</sub> (< 10 μmol O<sub>2</sub> kg<sup>-1</sup>) and low pH (Fig. 1), although the more pronounced OMZ at the Tehuantepec Bowl does appear to restrict the vertical distribution of some species.

### 4 Discussion

Our study of the vertical distribution of the cosomes is the first to describe the diel migration of four pteropod species into the pronounced OMZ of the ETP. In the hypoxic waters of the OMZ, these animals are surrounded by low pH water (7.4–7.5) and, based on

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the alkalinity of the region (~ 2350  $\mu$ mol kg<sup>-1</sup>, WOCE data P-18), are likely experiencing levels of CO<sub>2</sub> between 400–1000 ppm by a depth of 200 m, depending on latitude (Feely et al., 2004; Fabry et al., 2008; Byrne and Elliott, unpublished data). Since aragonite is thought to be undersaturated in this region, these results are surprising,

- <sup>5</sup> suggesting a greater resiliency in some species of pteropod with respect to acidification than has been previously inferred. Of the pteropods with habitats that include depths below 100 m, none responded to our hypercapnic treatment with a change in O<sub>2</sub> consumption or NH<sub>3</sub> excretion. *Diacria quadridentata* was the only species of tropical thecosome that was never found below the mixed layer, and it exhibited a respiratory
- <sup>10</sup> suppression and reduction of NH<sub>3</sub> excretion when exposed to hypercapnic conditions. In general, these results indicate that some species of pteropod will be able to function in the shallower, warmer, oxygenated end of their distribution under increasingly hypercapnic conditions. As calcifiers, these species may endure brief periods of acidosis by buffering their cellular pH though internal mechanisms or the dissolution of
- their aragonite shell. The mussel Mytilus galloprovincialis uses these tactics when exposed to elevated levels of CO<sub>2</sub>; they survive hypercapnia through decreased protein synthesis and shell dissolution (Michaelidis et al., 2005).

The natural exposure of diel vertically migratory pteropods to elevated  $CO_2$  would be under conditions of reduced temperature and  $O_2$  saturation, both of which have been shown to significantly depress the metabolic rate of a number of species (Seibel, 2011). This means that generally when pteropods in the ETP are experiencing elevated levels of  $CO_2$  their metabolic rate is already suppressed due to other environmental parameters. Exposure to hypercapnia independent of these other conditions may not be physiologically analogous to their response to hypercapnia at depth in the OMZ.

The migratory behavior of pteropods in the ETP, which regularly exposes individuals to elevated CO<sub>2</sub>, may have resulted in the development of physiological mechanisms by which these zooplankton can better cope with brief periods of hypercapnia (Seibel and Fabry, 2003; Seibel and Walsh, 2001, 2003). However, these distributional patterns and physiological studies do not rule out the possibility that ocean acidification

may have severe effects on pteropods. In fact, our results indicate that non-migratory species such as *Diacria quadridentata* could, in the absence of acclimation and adaptation, be significantly impacted by even brief periods of exposure to  $CO_2$  with unknown implications for species fitness, biogeography and survival. Furthermore, it has also

- <sup>5</sup> been shown that organisms capable of withstanding short periods of pronounced hypercapnia, such as the peanut worm, *Sipunculus nudus*, require time to restore extraand intracellular pH (Reipschlager and Pörtner, 1996; Langenbuch and Pörtner, 2002, 2004). Prolonged experience of high CO<sub>2</sub> resulted in death for *S. nudus*. Therefore, although capable of coping with 6–12 h of elevated levels of CO<sub>2</sub>, the perpetual acidifi-
- cation of surface waters may impact even the diel migratory plankton by compressing the regions of the ocean where recovery from acidosis is possible.

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### Table 1. Dates, locations and vessels used during sampling.

Date	Location	Vessel
Jun 2007	Gulf of California between 27° N 112° W and 111° W	R/V New Horizon
Oct-Nov 2007	Eastern Tropical Pacific between 9° N 90° W and 11° N 98° W	R/V Seward Johnson
Dec 2008–Jan 2009	Eastern Tropical Pacific between 9° N 90° W and 11° N 98° W	R/V Knorr

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**Table 2.** MOCNESS net parameters and average hydrographic data for each day and night net tow (tow identification, ID) at the Costa Rica Dome (CRD) and Tehuantepec Bowl (TB) during 2007 and 2008. Minimum and maximum pressures are recorded in decibars (dB) and served as a proxy for depth (1 dB  $\approx$  1 m). Volume of water filtered through each net was measured in m<sup>3</sup> (V.f.). MOCNESS data has been made available by K. Wishner.

CRD – Day						CRD – Night									
	20	07			200	08			20	07			200	08	
ID	Max	Min	V.f.	ID	Max	Min	V.f.	ID	Max	Min	V.f.	ID	Max	Min	V.f.
	(dB)	(dB)	(m <sup>3</sup> )		(dB)	(dB)	(m <sup>3</sup> )		(dB)	(dB)	(m <sup>3</sup> )		(dB)	(dB)	(m <sup>3</sup> )
616.4	400	350	736	637.4	400	350	751	615.4	400	350	429	641.5	400	350	808
616.5	350	300	385	637.5	350	300	867	615.5	350	300	588	641.6	350	300	935
616.6	300	250	452	637.6	300	250	785	615.6	300	250	389	641.7	300	250	764
616.7	250	200	405	637.7	250	200	815	615.7	250	200	515	641.8	250	200	834
616.8	200	150	686	637.8	200	150	645	615.8	200	150	370	-	200	150	-
618.1	150	100	731	635.1	150	100	552	621.1	150	100	517	638.1	150	100	484
618.2	100	80	457	635.2	100	80	267	621.2	100	80	244	638.2	100	80	288
618.3	80	60	349	635.3	80	60	334	621.3	80	60	383	638.3	80	60	329
618.4	60	50	229	635.4	60	50	211	621.4	60	50	180	638.4	60	50	186
618.5	50	40	431	635.5	50	40	168	621.5	50	40	147	638.5	50	40	214
618.6	40	30	282	635.6	40	30	98	621.6	40	30	232	638.6	40	30	188
618.7	30	20	273	635.7	30	20	248	621.7	30	20	93	638.7	30	20	238
618.8	20	0	397	635.8	20	0	330	621.8	20	0	398	638.8	20	0	238
TB – Day															
			TB –	Day							TB –	Night			
	20	07	TB –	Day	200	08		 	20	07	TB –	Night	200	08	
ID	20 Max	07 Min	TB – V.f.	Day	200 Max	08 Min	V.f.		20 Max	07 Min	TB – V.f.	Night ID	200 Max	08 Min	V.f.
ID	20 Max (dB)	07 Min (dB)	TB – V.f. (m <sup>3</sup> )	Day ID	200 Max (dB)	08 Min (dB)	V.f. (m <sup>3</sup> )		20 Max (dB)	07 Min (dB)	TB – V.f. (m <sup>3</sup> )	Night ID	200 Max (dB)	08 Min (dB)	V.f. (m <sup>3</sup> )
ID 611.3	20 Max (dB) 400	07 Min (dB) 350	TB – V.f. (m <sup>3</sup> ) 378	Day ID 630.4	200 Max (dB) 400	08 Min (dB) 350	V.f. (m <sup>3</sup> ) 738	ID 609.4	20 Max (dB) 550	07 Min (dB) 350	TB – V.f. (m <sup>3</sup> ) 1224	Night ID	200 Max (dB) 400	08 Min (dB) 350	V.f. (m <sup>3</sup> ) 655
ID 611.3 611.4	200 Max (dB) 400 350	07 Min (dB) 350 300	TB – V.f. (m <sup>3</sup> ) 378 1028	Day ID 630.4 630.5	200 Max (dB) 400 350	08 Min (dB) 350 300	V.f. (m <sup>3</sup> ) 738 659	ID 609.4 609.5	20 Max (dB) 550 350	07 Min (dB) 350 150	TB – V.f. (m <sup>3</sup> ) 1224 2035	Night ID 628.4 -	200 Max (dB) 400 350	08 Min (dB) 350 300	V.f. (m <sup>3</sup> ) 655
ID 611.3 611.4 611.5	200 Max (dB) 400 350 300	07 Min (dB) 350 300 250	TB – V.f. (m <sup>3</sup> ) 378 1028 474	Day ID 630.4 630.5 630.6	200 Max (dB) 400 350 300	08 Min (dB) 350 300 250	V.f. (m <sup>3</sup> ) 738 659 583	ID 609.4 609.5 612.1	20 Max (dB) 550 350 150	07 Min (dB) 350 150 100	TB – V.f. (m <sup>3</sup> ) 1224 2035 598	Night ID 628.4 - 628.6	200 Max (dB) 400 350 300	08 Min (dB) 350 300 250	V.f. (m <sup>3</sup> ) 655 – 606
ID 611.3 611.4 611.5 611.6	200 Max (dB) 400 350 300 250	07 Min (dB) 350 300 250 200	TB – V.f. (m <sup>3</sup> ) 378 1028 474 480	Day ID 630.4 630.5 630.6 630.7	200 Max (dB) 400 350 300 250	08 Min (dB) 350 300 250 200	V.f. (m <sup>3</sup> ) 738 659 583 763	ID 609.4 609.5 612.1 612.2	20 Max (dB) 550 350 150 100	07 Min (dB) 350 150 100 80	TB – V.f. (m <sup>3</sup> ) 1224 2035 598 360	Night ID 628.4 - 628.6 628.7	200 Max (dB) 400 350 300 250	08 Min (dB) 350 300 250 200	V.f. (m <sup>3</sup> ) 655 – 606 992
ID 611.3 611.4 611.5 611.6 606.1	200 Max (dB) 400 350 300 250 200	07 Min (dB) 350 300 250 200 100	TB – V.f. (m <sup>3</sup> ) 378 1028 474 480 1282	Day ID 630.4 630.5 630.6 630.7 630.8	200 Max (dB) 400 350 300 250 200	08 (dB) 350 300 250 200 150	V.f. (m <sup>3</sup> ) 738 659 583 763 502	ID 609.4 609.5 612.1 612.2 612.3	20 Max (dB) 550 350 150 100 80	07 Min (dB) 350 150 100 80 60	TB – V.f. (m <sup>3</sup> ) 1224 2035 598 360 392	Night ID 628.4 - 628.6 628.7 628.8	200 Max (dB) 400 350 300 250 200	08 (dB) 350 300 250 200 150	V.f. (m <sup>3</sup> ) 655 – 606 992 626
ID 611.3 611.4 611.5 611.6 606.1 606.2	200 Max (dB) 400 350 300 250 200 100	07 Min (dB) 350 300 250 200 100 80	TB – V.f. (m <sup>3</sup> ) 378 1028 474 480 1282 200	Day ID 630.4 630.5 630.6 630.7 630.8 626.1	200 Max (dB) 400 350 300 250 200 150	08 Min (dB) 350 300 250 200 150 100	V.f. (m <sup>3</sup> ) 738 659 583 763 502 481	ID 609.4 609.5 612.1 612.2 612.3 612.4	20 Max (dB) 550 350 150 100 80 60	07 Min (dB) 350 150 100 80 60 50	TB – V.f. (m <sup>3</sup> ) 1224 2035 598 360 392 327	Night ID 628.4 - 628.6 628.7 628.8 633.1	200 Max (dB) 400 350 300 250 200 150	08 Min (dB) 350 300 250 200 150 100	V.f. (m <sup>3</sup> ) 655 - 606 992 626 749
ID 611.3 611.4 611.5 611.6 606.1 606.2 606.3	200 Max (dB) 400 350 300 250 200 100 80	07 Min (dB) 350 300 250 200 100 80 60	TB – V.f. (m <sup>3</sup> ) 378 1028 474 480 1282 200 233	Day ID 630.4 630.5 630.6 630.7 630.8 626.1 626.2	200 Max (dB) 400 350 300 250 200 150 100	08 Min (dB) 350 300 250 200 150 100 80	V.f. (m <sup>3</sup> ) 738 659 583 763 502 481 431	ID 609.4 609.5 612.1 612.2 612.3 612.4 612.5	20 Max (dB) 550 350 150 100 80 60 50	07 Min (dB) 350 150 100 80 60 50 40	TB – V.f. (m <sup>3</sup> ) 1224 2035 598 360 392 327 256	Night ID 628.4  628.6 628.7 628.8 633.1 633.2	200 Max (dB) 400 350 300 250 200 150 100	08 Min (dB) 350 300 250 200 150 100 80	V.f. (m <sup>3</sup> ) 655 - 606 992 626 749 333
ID 611.3 611.4 611.5 611.6 606.1 606.2 606.3 606.4	200 Max (dB) 400 350 300 250 200 100 80 60	07 Min (dB) 350 250 200 100 80 60 50	TB - V.f. (m <sup>3</sup> ) 378 1028 474 480 1282 200 233 81	Day ID 630.4 630.5 630.6 630.7 630.8 626.1 626.2 626.3	200 Max (dB) 400 350 300 250 200 150 100 80	08 Min (dB) 350 250 200 150 100 80 60	V.f. (m <sup>3</sup> ) 738 659 583 763 502 481 431 503	ID 609.4 609.5 612.1 612.2 612.3 612.4 612.5 612.6	20 Max (dB) 550 350 150 100 80 60 50 40	07 Min (dB) 350 150 100 80 60 50 40 30	TB - V.f. (m <sup>3</sup> ) 1224 2035 598 360 392 327 256 400	Night ID 628.4 - 628.6 628.7 628.8 633.1 633.2 633.3	200 Max (dB) 400 350 300 250 200 150 100 80	08 Min (dB) 350 300 250 200 150 100 80 60	V.f. (m <sup>3</sup> ) 655 - 606 992 626 749 333 412
ID 611.3 611.4 611.5 611.6 606.1 606.2 606.3 606.4 606.5	200 Max (dB) 400 350 300 250 200 100 80 60 50	07 Min (dB) 350 250 200 100 80 60 50 40	TB - V.f. (m <sup>3</sup> ) 378 1028 474 480 1282 200 233 81 118	Day ID 630.4 630.5 630.6 630.6 630.7 630.8 626.1 626.2 626.3 626.4	200 Max (dB) 400 350 300 250 200 150 100 80 60	08 Min (dB) 350 250 200 150 100 80 60 50	V.f. (m <sup>3</sup> ) 738 659 583 763 502 481 431 503 241	ID 609.4 609.5 612.1 612.2 612.3 612.4 612.5 612.6 612.7	20 Max (dB) 550 350 150 100 80 60 50 40 30	07 Min (dB) 350 150 100 80 60 50 40 30 20	TB - V.f. (m <sup>3</sup> ) 1224 2035 598 360 392 327 256 400 291	Night ID 628.4 - 628.7 628.8 633.1 633.2 633.3 633.4	200 Max (dB) 400 350 300 250 200 150 100 80 60	08 Min (dB) 350 250 200 150 100 80 60 50	V.f. (m <sup>3</sup> ) 655 - 606 992 626 749 333 412 102
ID 611.3 611.4 611.5 611.6 606.1 606.2 606.3 606.4 606.5 606.6	200 Max (dB) 400 350 300 250 200 100 80 60 50 40	07 Min (dB) 350 300 250 200 100 80 60 50 40 30	TB - V.f. (m <sup>3</sup> ) 378 1028 474 480 1282 200 233 81 118 108	Day ID 630.4 630.5 630.6 630.6 630.7 630.8 626.1 626.2 626.3 626.4 626.5	200 Max (dB) 400 350 300 250 200 150 100 80 60 50	08 Min (dB) 350 300 250 200 150 100 80 60 50 40	V.f. (m <sup>3</sup> ) 738 659 583 763 502 481 431 503 241 201	ID 609.4 609.5 612.1 612.2 612.3 612.4 612.5 612.6 612.7 612.8	20 Max (dB) 550 350 150 100 80 60 50 40 30 20	07 Min (dB) 350 150 100 80 60 50 40 30 20 0	TB - V.f. (m <sup>3</sup> ) 1224 2035 598 360 392 327 256 400 291 535	Night ID 628.4 - 628.6 628.7 628.8 633.1 633.2 633.3 633.4 633.5	200 Max (dB) 400 350 300 250 200 150 100 80 60 50	08 Min (dB) 350 300 250 200 150 100 80 60 50 40	V.f. (m <sup>3</sup> ) 655 - 606 992 626 749 333 412 102 144
ID 611.3 611.4 611.5 611.6 606.1 606.2 606.3 606.4 606.5 606.6 606.7	200 Max (dB) 400 350 300 250 200 100 80 60 50 40 30	07 Min (dB) 350 300 250 200 100 80 60 50 40 30 20	TB - V.f. (m <sup>3</sup> ) 378 1028 474 480 1282 200 233 81 118 108 78	Day ID 630.4 630.5 630.6 630.7 630.8 626.1 626.2 626.3 626.4 626.5 626.6	200 Max (dB) 400 350 300 250 200 150 100 80 60 50 40	08 Min (dB) 350 300 250 200 150 150 100 80 60 50 40 30	V.f. (m <sup>3</sup> ) 738 659 583 763 503 481 431 503 241 201 152	ID 609.4 609.5 612.1 612.2 612.3 612.4 612.5 612.6 612.7 612.8 -	20 Max (dB) 550 350 150 100 80 60 50 40 30 20	07 Min (dB) 350 150 100 80 60 50 40 30 20 0 -	TB - V.f. (m <sup>3</sup> ) 1224 2035 598 360 392 327 256 400 291 535 -	Night ID 628.4 - 628.6 628.7 628.8 633.1 633.2 633.3 633.4 633.5 633.6	200 Max (dB) 400 350 300 2500 200 150 100 80 60 50 40	08 Min (dB) 350 250 200 150 100 80 60 50 40 30	V.f. (m <sup>3</sup> ) 655 606 992 626 749 333 412 102 144 211
ID 611.3 611.4 611.5 611.6 606.1 606.2 606.3 606.4 606.5 606.6 606.7 606.8	200 Max (dB) 400 350 250 200 100 80 60 50 40 30 20	07 Min (dB) 350 300 250 200 100 80 60 50 40 30 20 0	TB - V.f. (m <sup>3</sup> ) 378 1028 474 480 1282 200 233 81 118 108 78 300	Day ID 630.4 630.5 630.6 630.7 630.8 626.1 626.2 626.3 626.4 626.5 626.6 626.7	200 Max (dB) 400 350 300 250 200 150 100 80 60 50 40 30	08 Min (dB) 350 250 200 150 100 80 60 50 40 30 20	V.f. (m <sup>3</sup> ) 738 659 583 763 503 503 241 201 152 164	ID 609.4 609.5 612.1 612.2 612.3 612.4 612.5 612.6 612.7 612.8 –	20 Max (dB) 550 350 150 100 80 50 40 30 20 -	07 Min (dB) 350 150 100 80 60 50 40 30 20 0 -	TB - V.f. (m <sup>3</sup> ) 1224 2035 598 360 392 327 256 400 291 535 -	Night ID 628.4 - 628.6 628.7 628.8 633.1 633.2 633.4 633.5 633.6 633.6 633.7	200 Max (dB) 400 350 300 250 250 250 150 150 100 80 60 50 40 30	08 Min (dB) 350 250 200 150 100 80 60 50 40 30 20	V.f. (m <sup>3</sup> ) 655 606 992 626 749 333 412 102 144 211 177

	0	.03 % CC	) <sub>2</sub>	0.10 % CO <sub>2</sub>			
	а	b	r <sup>2</sup>	a	b	r <sup>2</sup>	
Cavolinia longirostris	0.23	-0.75	0.88	0.01	-1.38	0.83	
Clio pyramidata	0.27	-0.72	0.67	0.16	-0.81	0.93	
Hyalocylis striata	0.27	-0.69	0.53	0.27	-0.64	0.43	
Diacria quadridentata	0.6	-0.58	0.41	0.01	-1.26	0.98	
Creseis virgula	0.39	-0.56	0.88	0.13	-0.78	0.99	

**Table 3.** Scaling curves describing the relationship between wet mass (*M*, g) and oxygen consumption rate (*R*,  $\mu$ mol O<sub>2</sub> g<sup>-1</sup> h<sup>-1</sup>) following the relationship *R* = *aM*<sup>*b*</sup> (Fig. 2).

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		Mean wet weight (mg)	MO <sub>2</sub>	2 (μmol O <sub>2</sub> g <sup>-1</sup> h <sup>-1</sup> ) Mean	MNH n	<sub>3</sub> (μmol NH <sub>3</sub> g <sup>-1</sup> h <sup>-1</sup> ) Mean	n	O:N Mean
Hvalocvlis striata	control	10.5±4.2	19	7.31±3.64	17	0.55±0.44	17	40.3±25.7
,,	0.10 % CO <sub>2</sub>	$8.4 \pm 4.4$	16	$7.07 \pm 3.57$	14	$0.33 \pm 0.16$	14	$44.4 \pm 17.9$
Creseis virgula	control	6.8±3.8	10	$7.75 \pm 4.17$	8	$0.52 \pm 0.15$	8	$35.3 \pm 17.7$
	0.10 % CO <sub>2</sub>	6.3±2.3	3	$7.20 \pm 2.38$	1	0.75	1	14.5
Clio pyramidata	control	9.1±4.9	13	$9.96 \pm 4.80$	10	$0.70 \pm 0.47$	10	$31.8 \pm 8.3$
	0.10 % CO <sub>2</sub>	$13.5 \pm 6.9$	9	$8.55 \pm 7.80$	8	$0.83 \pm 0.89$	8	$25.0 \pm 5.0$
Cavolinia longirostris	control	8.2±3.7	20	$12.29 \pm 7.60$	20	1.21±0.65	20	$20.5 \pm 8.6$
	0.10 % CO <sub>2</sub>	$5.9 \pm 2.6$	18	$12.82 \pm 7.45$	18	$1.32 \pm 0.70$	18	$22.2 \pm 11.0$
Diacria quadridentata	control	9.7±3.3	12	$10.62 \pm 5.63$	11	0.89±0.44	11	$27.5 \pm 12.9$
	0.10 % CO <sub>2</sub>	10.9±6.2	7	$5.01 \pm 4.73$	6	$0.33 \pm 0.16$	6	$30.3 \pm 28.6$

**Table 4.** The average size, oxygen consumption ( $MO_2 \pm SD$ ), and ammonia excretion ( $MNH_3 \pm SD$ ) of the cosome pteropods found in the Eastern Tropical Pacific at 20 °C.

Table 5. Statistical analysis comparing the difference between groups exposed to 0.03 % and 0.10% CO<sub>2</sub> (Table 4). Statistical analysis for O<sub>2</sub> consumption (dependant variable) was conducted using a one-way ANCOVA to account for the variation due to size (continuous variable). Analysis of NH<sub>3</sub> excretion and O: N ratio was conducted using a two-tailed t-test.

Species	MO <sub>2</sub>	MNH <sub>3</sub>	0:N
Hyalocylis striata	$F_{(1.43)} = 1.03, p = 0.315$	$t_{42} = 0.064, p = 0.948$	$t_{42} = -0.853, p = 0.399$
Cavolinia longirostris	$F_{(1,35)} = 3.75, p = 0.061$	$t_{36} = 0.494, p = 0.624$	$t_{36} = 0.535, p = 0.596$
Creseis virgula	$F_{(1,10)} = 0.20, p = 0.660$	$t_7 = -1.026, p = 0.339$	$t_7 = 1.032, p = 0.336$
Clio pyramidata	$F_{(1,19)} = 1.48, p = 0.240$	$t_{16} = 0.415, p = 0.683$	$t_{17} = 2.047, p = 0.057$
Diacria quadridentata	$F_{(1,16)} = 5.45, p = 0.033$	$t_{15} = 2.975, p = 0.009$	$t_{15} = -0.290, p = 0.776$





Fig. 1. A typical temperature (°C, black) and  $O_2$  (µmol kg<sup>-1</sup>, grey) profile for the Gulf of California in June 2007 (A), the Tehuantepec Bowl (B), and the Costa Rica Dome (C). The profiles from the Tehuantepec Bowl and the Costa Rica Dome are from 2008 and also show a pH profile (dashed black).

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**Fig. 4.** The average  $NH_3$  excretion for normocapnic (0.03%, white) and hypercapnic (0.10%, black) treatments (Table 4). Error bars represent one standard deviation from the mean. Only one data point was available for the species *Creseis virgula* in the hypercapnic treatment. *Diacria quadridentata* was the only the cosome which responded with a significant change in  $NH_3$  excretion (\*).

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**Fig. 5.** The day (grey bar) and night (black bar) location of tropical pteropod species is plotted alongside the O<sub>2</sub> concentration at depth for 2007 and 2008 (solid dark grey line,  $\mu$ mol kg<sup>-1</sup>) and pH in 2008 (dashed dark grey line). O<sub>2</sub> profiles are from MOCNESS data (assembled by R. Williams, D. Outram and K. Wishner), and pH data are from the MICA system of Byrn and Elliot (unpublished data).

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